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## (54) CANCER REMEDY COMPRISING ANTHRANILIC ACID DERIVATIVE AS ACTIVE INGREDIENT

(57) A cancer remedy comprising a compound represented by the following formula as an active ingredient:

wherein, X represents a group represented by either of the following formulae:

$$R^5$$
 $R^2$ 
 $L^1$ 

and

wherein,  $R^1$  and  $R^2$  represent each a hydrogen atom, a hydroxy group, a trihalomethyl group, a  $C_1$ - $C_{12}$  alkoxy group or alkylthio group, a (substituted)  $C_7$ - $C_{11}$  aralkyloxy group or a (substituted)  $C_3$ - $C_{10}$  alkenyloxy group;  $R^4$  and  $R^5$  represent each a hydrogen atom, a halogen atom, a  $C_1$ - $C_4$  alkyl group or a  $C_1$ - $C_4$  alkoxy group; A represents -O-, -S-, -S (=O)-, -S(=O)<sub>2</sub>-, -CH<sub>2</sub>-, -OCH<sub>2</sub>-, -SCH<sub>2</sub>, -C(=O)- or -CH(OR<sup>6</sup>)-; Y represents a hydrogen atom, a halogen atom, a nitro

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group, a nitrile group, an amino group, -COOR $^7$ , -NHCOR $^8$  or -NHSO $_2$ R $^9$ ; E represents -C(=O)-, -CR $^{10}$ R $^{11}$ C(=O)-, -CH $_2$ CH $_2$ C(=O)- or -CH=CHC(=O)-; G represents a hydrogen atom, a hydroxy group, -SO $_2$ NH $_2$ , -COOR $^3$ , -CN or a tetrazol-5-yl group; and Z represents a hydrogen atom, a halogen atom, a nitro group or a methyl group.

#### Description

**Technical Field** 

[0001] The present invention relates to a cancer remedy comprising an anthranilic acid derivative or a pharmaceutically acceptable salt as an active ingredient. More particularly, it relates to a cancer remedy comprising an anthranilic acid derivative, having an anthranilic acid skeleton and a benzene skeleton and further having the benzene skeleton or a naphthalene skeleton, or a pharmaceutically acceptable salt thereof, as an active ingredient.

## 10 Background Art

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[0002] There have been strong demands from the society for the development of excellent carcinostatic agents, and it is extremely important to create a new compound having strong cytotoxicity in development of excellent carcinostatic agents. In general, the carcinostatic activity of. compounds and the carcinostatic spectra depend largely on the chemical structure thereof. Accordingly, there is a very great possibility of developing better carcinostatic agents than those put into practical use at present from cytotoxic compounds having a novel structure.

[0003] The carcinostatic activity based on the cytotoxic activity of compounds having an aryl skeleton are known as, for example substituted phenylsulfonyl derivatives [JP-A No. 5-9170 (hereinafter, JP-A refers to Japanese Unexamined Patent Publication)] 2-arylquinolinol derivatives (JP-A No. 7-33743) and benzoylacetylene derivatives (JP-A No. 7-97350).

[0004] On the other hand, WO95/32943 and Journal of Medicinal Chemistry (J. Med. Chem.), vol. 40, No.4, pp. 395-407 (1997) describe compounds having a naphthalene skeleton and an anthranilic acid skeleton and an antiallergic activity and an inhibitory activity against the production of IgE antibodies.

[0005] WO97/19910 describes compounds having a benzene skeleton and an anthranilic acid skeleton and further an antiallergic activity and an inhibitory activity against the production of IgE antibodies.

[0006] However, it is unknown that a group of compounds having the aryl skeleton and the anthranilic acid skeleton at the same time have a cytotoxic activity or a carcinostatic activity.

#### Disclosure of the Invention

[0007] An object of the present invention is to provide a cancer remedy having a novel structure.

[0008] About the problems described above, the inventors of the present application have newly found that the anthranilic acid derivatives have a cytotoxic activity against cell strains having a high growth property, and that the anthranilic acid derivatives have a strong growth inhibitory activity or cytotoxic activity against human cancer cells. Therefore, a pharmaceutical composition comprising the anthranilic acid derivative, its pharmaceutically acceptable salt or a pharmaceutically acceptable solvate thereof as an active ingredient is useful as a cancer remedy.

[0009] Namely, the present invention provides a cancer remedy comprising an anthranilic acid derivative represented by the following formula (1) or a pharmaceutically acceptable salt thereof as an active ingredient:

$$X \cdot A \longrightarrow F \cdot N \longrightarrow Z$$
(1)

wherein, X represents a group selected from the following formula (2)-1 and formula (2)-2 in the formula (1):

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$$R^1$$
 (2) -1 and  $R^2$  (2) -2

wherein,  $R^1$  and  $R^2$  represent each independently a hydrogen atom, a hydroxy group, a trihalomethyl group, an alkoxy group or an alkylthio group comprising a  $C_1$ - $C_{12}$  chain or cyclic hydrocarbon group and an oxy group or a thio group, a  $C_7$ - $C_{11}$  aralkyloxy group wherein an aryl group moiety may be substituted with one or more halogen atoms, methyl groups or methyloxy groups or a  $C_3$ - $C_{10}$  alkenyloxy group which may be substituted with one or more phenyl groups;  $R^4$  and  $R^5$  represent each independently a hydrogen atom, a halogen atom, a  $C_1$ - $C_4$  alkyl group or a  $C_1$ - $C_4$  alkoxy group, in the formula (2)-1 or the formula (2)-2;

A represents a bond; -O-, -S-, -S(=O)-, -S(=O) $_2$ -, -CH $_2$ -, -OCH $_2$ -, -SCH $_2$ -, -C(=O)- or -CH(OR $^6$ )-, wherein, R $^6$  represents a hydrogen atom or a C $_1$ -C $_4$  alkyl group;

Y represents a hydrogen atom, a halogen atom, a nitro group, a nitrile group, an amino group, -COOR<sup>7</sup>, -NHCOR<sup>8</sup> or -NHSO<sub>2</sub>R<sup>9</sup>,

wherein,  $R^7$  represents a hydrogen atom or a  $C_1$ - $C_4$  alkyl group;  $R^8$  and  $R^9$  represent each independently a  $C_1$ - $C_4$  alkyl group;

E represents a bond; -C(=O)-, -CR<sup>10</sup>R<sup>11</sup>C(=O)- (wherein, R<sup>10</sup> and R<sup>11</sup> represent each independently a hydrogen atom or a fluorine atom),

-CH<sub>2</sub>CH<sub>2</sub>C(=O)- or -CH=CHC(=O)-;

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G represents a hydrogen atom, a hydroxy group,  $-SO_2NH_2$ ,  $-COOR^3$ , (wherein,  $R^3$  represents a hydrogen atom or a  $C_1$ - $C_4$  alkyl group), -CN or a tetrazol-5-yl group; and

Z represents a hydrogen atom, a halogen atom, a nitro group or a methyl group.

[0010] Furthermore, the present invention provides a therapy for cancer using a drug comprising the anthranilic acid derivative or a pharmaceutically acceptable salt thereof.

[0011] In addition, the present invention is the use of the anthranilic acid derivative or a pharmaceutically acceptable salt thereof in order to produce the cancer remedy.

## **Best Mode for Carrying Out the Invention**

[0012] In the formula (2)-1 or (2)-2 in the above formula (1) representing the anthranilic acid derivative used in the present invention, R¹ and R² represent each independently a hydrogen atom, a hydroxy group, a trihalomethyl group, an alkoxy group or an alkylthio group comprising a C₁-C₁₂ chain or cyclic hydrocarbon group and an oxy group or a thio group, a C₂-C₁₁ aralkyloxy group wherein an aryl group moiety may be substituted with one or more atoms, methyl groups or methyloxy groups or a C₃-C₁₀ alkenyloxy group which may be substituted with one or more phenyl groups.

[0013] When R¹ or R² represents a C₁-C₁₂ chain or cyclic alkyloxy group, R¹ or R² can be selected from, for example methyloxy group, ethyloxy group, propyloxy group, 2-propyloxy group, (1- or 2-)methylpropyloxy group, 2,2-dimethylpropyloxy group, (n- or tert-)butyloxy group, 2-ethylbutyloxy group, (2- or 3-)methylbutyloxy group, pentyloxy group, hexyloxy group, octyloxy group, decyloxy group, dodecyloxy group, cyclopropyloxy group, cyclopentyloxy group, cyclohexyloxy group, cyclohexylmethyloxy group, cyclohexyloxy group, cyclohe

[0014] When  $R^1$  or  $R^2$  represents a  $C_1$ - $C_{12}$  chain or cyclic alkylthio group,  $R^1$  or  $R^2$  can be selected from, for example methylthio group, ethylthio group, propylthio group, 2-propylthio group, (1- or 2-)methylpropylthio group, 2,2-dimethylpropylthio group, (n- or tert-)butylthio group, 2-ethylbutyltyhio group, (2- or 3-)methylbutylthio group, pentylthio group, hexylthio group, octylthio group, decylthio group, dodecylthio group, cyclopropylthio group, cyclopentylthio group, cyclopentylthio group, cyclohexylmethylthio group, cyclohexylmethylthio group, cyclohexylmethylthio group, cyclohexylthio group, cyclohexylth

[0015] When  $R^1$  or  $R^2$  represents a  $C_7$ - $C_{12}$  aralkyloxy group, the aryl group moiety of the aralkyloxy group may be represented by one or more of halogen atoms, methyl groups or methyloxy groups, and examples of the substituents include fluorine atoms, chlorine atoms, bromine atoms, methyl groups, methyloxy groups and the like. Therefore, the aralkyloxy groups represented by  $R^1$  can be selected from, for example benzyloxy group, (2-, 3- or 4-)chlorobenzyloxy group, (2-, 3- or 4-)methoxybenzyloxy group, (2-, 3- or 4-)methylbenzyloxy group, ( $\alpha$  - or  $\beta$  -)phenethyloxy group, 3-phenylpropyloxy group, 2-phenyl-2-propyloxy group, 2-phenyl-1-cyclohexyloxy group, (1-phenylcyclopropyl)methyl-

loxy group, (1-phenylcyclopentyl)methyloxy group, (1- or 2-)naphthylmethyloxy group and the like.

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**[0016]** Furthermore, R¹ or R² may be a C₃-C₁₀ alkenyloxy group. The alkenyloxy group in this case can be selected from, for example allyloxy group, methallyloxy group, crotyloxy group, 3-butenyloxy group, 4-pentenyloxy group, 5-hexenyloxy group, 7-octenyloxy group, geranyloxy group, cinnamyloxy group, 2-cyclohexenyloxy group, (3-cyclohexenyl) methyloxy group, 1,4-pentadien-3-yloxy group and the like.

[0017] R¹ and R² may each be a hydrogen atom, a hydroxy group or a trihalomethyl group. Examples of the halogen atom representing the trihalomethyl group include fluorine atoms, chlorine atoms and the like.

[0018] Examples of preferable atoms or groups among the atoms or groups represented by  $R^1$  or  $R^2$  include a hydrogen atom, a hydroxy group, methyloxy group, ethyloxy group, propyloxy group, 2-propyloxy group, (1-or 2-)methylpropyloxy group, 2,2-dimethylpropyloxy group, (n- or tert)butyloxy group, 2-ethylbutyloxy group, (2- or 3-)methylbutyloxy group, pentyloxy group, hexyloxy group, heptyloxy group, octyloxy group, decyloxy group, dodecyloxy group, cyclopropylmethyloxy group, cyclopentyloxy group, cyclohexyloxy group, cyclohexylmethyloxy group, cyclooctyloxy group, cycloheptyloxy group, cyclodecyloxy group, methythio group, ethylthio group, isopropylthio group, t-butylthio group, 3-pentylthio group, cyclohexylthio group, group, 4-chlorobenzyloxy group, 4-methylbenzyloxy group, ( $\alpha$ - or  $\beta$ -)phenethyloxy group, 3-phenylpropyloxy group, (1- or 2-)naphthylmethyloxy group, allyloxy group, 3-butenyloxy group, 4-pentenyloxy group, 5-hexenyloxy group, 7-octenylxy group, trifluoromethyl group and the like.

[0019] Among them, the atoms or groups are especially preferably an alkyloxy group wherein the R¹ or R² group comprises a hydrogen atom, a hydroxy group, a C¹-C¹2 chain or cyclic saturated hydrocarbon or a C²-C¹2 aralkyloxy group, more preferably a hydrogen atom, for example a C⁵-C¹2 cyclic saturated alkoxy group such as cyclohexyloxy group, cycloheptyloxy group, cyclopentyloxy group or cyclododecanyloxy group or a C³-C8 branched chain saturated alkoxy group, especially preferably an alkyloxy group producing a branch from the adjacent carbon of the oxygen atom, for example isopropyloxy group, 3-pentyloxy group or benzyloxy group.

[0020] In the above formula (2)-1 or (2)-2, R¹ and R² may be substituted at an optional position on the naphthalene ring or the benzene ring; however, R¹ is preferably located in the 6-position counted from the ring in which the A on the naphthalene ring is linked (A is substituted at the 2-position) or R² is preferably located in the 4-position counted from the bond of A on the benzene ring.

[0021] In the above formula (2)-2,  $R^4$  and  $R^6$  represent each independently a hydrogen atom, a halogen atom, a  $C_1$ - $C_4$  alkyl group or a  $C_1$ - $C_4$  alkoxy group. Examples of the halogen atom include a fluorine atom, a chlorine atom, a bromine atom and the like. Examples of the  $C_1$ - $C_4$  alkyl group include methyl group, ethyl group, isopropyl group, t-butyl group and the like. Further, examples of the  $C_1$ - $C_4$  alkoxy group include methyloxy group, ethyloxy group, isopropyloxy group, t-butyloxy group and the like. Among the groups, examples of  $R^4$  or  $R^5$  include preferably a hydrogen atom, a chlorine atom, methyl group or methyloxy group. Among them, hydrogen atom is preferable. In addition,  $R^4$  is substituted at the 2-position on the benzene ring, and  $R^5$  is substituted at the 3-position on the benzene ring.

[0022] In the formula (1), A represents a bond; -O-, -S-, -S(=O)-, -S(=O)<sub>2</sub>-, -CH<sub>2</sub>-, -OCH<sub>2</sub>-, -SCH<sub>2</sub>-, -C(=O)- or -CH (OR<sup>6</sup>)-, wherein, R<sup>6</sup> represents a hydrogen atom or a C<sub>1</sub>-C<sub>4</sub> alkyl group.

Examples of the  $C_1$ - $C_4$  alkyl group include methyl group, ethyl group, n-propyl group, tert-butyl group and the like.  $R^6$  is preferably a hydrogen atom or methyl group. More preferable bond is -O- or -S- as A.

[0023] Furthermore, in the formula (1), Y represents a hydrogen atom, a halogen atom, a nitro group, a nitrile group, an amino group, -COOR<sup>7</sup>, wherein, R<sup>7</sup> represents a hydrogen atom or a C<sub>1</sub>-C<sub>4</sub> alkyl group, -NHCOR<sup>8</sup> or -NHSO<sub>2</sub>R<sup>9</sup>, wherein, R<sup>8</sup> and R<sup>9</sup> represent each independently a C<sub>1</sub>-C<sub>4</sub> alkyl group. When Y represents a halogen atom, examples of the halogen atom include a fluorine atom, a chlorine atom and a bromine atom. Among them, chlorine atom is preferable. When Y represents -COOR<sup>7</sup>, R<sup>7</sup> represents a hydrogen atom or a C<sub>1</sub>-C<sub>4</sub> alkyl group, and examples thereof include a hydrogen atom, methyl group, ethyl group, n-propyl group, 2-propyl group, n-butyl group and tert-butyl group. It is preferable for the -COOR<sup>7</sup> to be -COOH or -COOCH<sub>3</sub>. When Y represents -NHCOOR<sup>8</sup> or -NHSO<sub>2</sub>R<sup>9</sup>, R<sup>8</sup> or R<sup>9</sup> represents a C<sub>1</sub>-C<sub>4</sub> alkyl group. Examples thereof include methyl group, ethyl group, n-propyl group, 2-propyl group, n-butyl group and tert-butyl group. When Y represents -NHCOR<sup>8</sup> or -NHSO<sub>2</sub>R<sup>9</sup>, it is preferable for Y to be -NHCOCH<sub>3</sub> or -NHSO<sub>2</sub>CH<sub>3</sub>.

[0024] It is especially preferable for Y to be a hydrogen atom, a chlorine atom, a nitro group or a nitrile group, among ones listed above.

[0025] Furthermore, in the above formula (1), E represents a bond; -C(=O)-, -CR<sup>10</sup>R<sup>11</sup>C(=O)-, wherein R<sup>10</sup> and R<sup>11</sup> represent each independently a hydrogen atom or a fluorine atom,

-CH<sub>2</sub>CH<sub>2</sub>C(=O)- or -CH=CHC(=O)-. Among them, E represents preferably a bond; -C(=O)- or -CH<sub>2</sub>C(=O)-, more preferably a bond; -CH<sub>2</sub>C(=O)-.

[0026] In the above formula (1), G represents a hydrogen atom, a hydroxy group, -SO<sub>2</sub>NH<sub>2</sub>, -COOR<sup>3</sup>, wherein, R<sup>3</sup> represents a hydrogen atom or a C<sub>1</sub>-C<sub>4</sub> alkyl group,

-CN or a tetrazol-5-yl group. When G represents -COOR<sup>3</sup>, examples of the alkyl group represented by R<sup>3</sup> include methyl group, ethyl group, (n- or iso)propyl group, (n-, iso- or tert-)butyl group and the like. The G is preferably -COOR<sup>3</sup>,

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wherein  $R^3$  represents a hydrogen atom or a  $C_1$ - $C_4$  alkyl group, or tetrazol-5-yl group; and  $R^3$  is especially preferably a hydrogen atom, methyl group or ethyl group. It is more preferable for G to be -COOH or tetrazol-5-yl group.

[0027] In addition, in the above formula (1), Z represents a hydrogen atom, a halogen atom, a nitro group or a methyl group. Examples of the halogen atom include a fluorine atom, a chlorine atom and a bromine atom. Z is preferably a hydrogen atom, a fluorine atom, a chlorine atom and methyl group, especially preferably a hydrogen atom.

**[0028]** In the anthranilic acid derivative represented by the formula (1), it is especially preferable that  $R^1$  and  $R^2$  represent each a hydrogen atom, a  $C_1$ - $C_{12}$  alkoxy group or a  $C_7$ - $C_{12}$  aralkyloxy group; A represents a bond; -O-; and E represents a bond; -C(=O)- or -CH<sub>2</sub>C(=O)-. The compound manifests an exceedingly high cytotoxic activity against cells having a strong growth activity.

[0029] In the anthranilic acid derivative represented by the formula (1), it is preferable that R¹ and R² represent each a hydrogen atom, a C<sub>5</sub>-C<sub>12</sub> cyclic alkyl group, a C<sub>3</sub>-C<sub>8</sub> branched chain alkyl group or a benzyl group; A represents a bond; -O-; E represents a bond; -CH<sub>2</sub>C(=O)-; and G represents COOH or a tetrazol-5-yl group. The compound manifests a stronger cytotoxic activity.

[0030] In the above formula (1), when Z represents a halogen atom or a methyl group, the substituent is preferably located in the 4- or the 5-position with respect to the group G on the benzene ring to which the group Z represents bound. The Z group located in the 4- or the 5-position has advantages in preventing the compound represented by the formula (1) from being inactivated with metabolism and sustaining the pharmaceutical effects thereof.

[0031] When Y is -COOH group or G is a -COOH group or a tetrazol-5-yl group in the above formula (1) (Y and G may be present at the same time or only either one thereof may be present), the carboxylic acid group or the like, if necessary, may be converted into a pharmaceutically acceptable nontoxic salt thereof. Examples of the preferably used nontoxic salt-forming cation include alkali metal ions such as Na+ and K+; alkaline earth metal ions such as Mg<sup>2+</sup> and Ca<sup>2+</sup>; nontoxic equivalent metal ions such as Al<sup>3+</sup> and Zn<sup>2+</sup>; organic bases such as ammonia, triethylamine, ethylenediamine, propanediamine, pyrrolidine, piperidine, piperazine, pyridine, lysine, choline, ethanolamine, N,N-dimethylethanolamine, 4-hydroxypiperidine, glucosamine, N-methylglucamine and the like. Among the salt-forming cation, Na<sup>2+</sup>, Ca<sup>2+</sup> and the organic bases such as lysine, choline, N,N-dimethylethanolamine and N-methylglucamine are preferably used.

[0032] The anthranilic acid derivative or a nontoxic salt thereof represented by the formula (1) may form a pharmaceutically acceptable solvate thereof. Solvents forming the solvate can be selected from water, methanol, ethanol, (n-and iso-)propyl alcohol, (n- and tert-)butanol, acetonitrile, acetone, methyl ethyl ketone, chloroform, ethyl acetate, diethyl ether, tert-butyl methyl ether, benzene, toluene, DMF, DMSO and the like. Among the solvents, water, methanol, ethanol, (n- and iso-)propyl alcohol or acetonitrile is especially preferably used.

[0033] The cancer remedy of the present invention comprises the anthranilic acid derivative, a pharmaceutically acceptable salt thereof or a solvate thereof as an active ingredient; however, a pharmaceutically acceptable carrier, if necessary, may be added.

[0034] Preferable specific examples of the anthranilic acid derivative represented by the formula (1) are listed in the following tables. When the structural formula of the compound has an asymmetric carbon (for example, compound No. 44), all the optical isomers are included. When the structural formula has a carbon-carbon double bond (for example, compound No. 120), both geometrical isomers are included. In the tables, "tet" indicates a tetrazol-5-yl group.

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	4	<del>Щ</del> 2	<b>8</b>	С <del>1</del>	뜐	ජි	중	ਤੰ	ર્કે	OCH <sub>2</sub>
30	R5	١	i	١	I	1	1	1	1	1
35		1	. 1	. 1	. 1	1	ł	. 1	i	. 1
	Position	I	ı	I	1	1	1	1	ł	1
40	R1 and R2	Ŧ	Ŧ	£	±	±	Ŧ	Ŧ.	±	±
45	RI									
50	×	Naphtha lene	Naphthalene	Naphthalene	Naphthalene	Naphtha ene	Naphthalene	Naphtha i ene	Naphtha i ene	Naphtha   ene
55	Compound No.	28	59	30	31	32	æ	34	35	36

5	Position		1	I	1	ı	ſ	ı	1	ı
	N	=	I	<b>±</b>	Ξ	×	I	×	=	±
10	Ø	H000	СООМе	H000	СООМе	C00H	СООМе	Н000	СООМе	COOH
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	පි	8	8	8	8	8	8	8	8
<i>25</i>	>	王	=	<b>=</b>	x	<b>=</b>	=	=	<b>=</b>	=
	∢	0CH <sub>2</sub>	SCH <sub>2</sub>	SCH <sub>2</sub>	8	9	03	8	CH (OMe)	CH (OMe)
30	R5	1	1	1	1	ì	1	1	1	1
35	R4	1	1	1	ŧ	1	1	1	1	1
	Position	·	ı	1	t	1	ı	1	1	1
40	Ri and R2	±	Ŧ	Ŧ	±	±	Ŧ	Ŧ		土
45	RI a	_	-		_		_	_		
50	×	Naphthalene	Naphtha ene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphtha lene	Naphthalene	Naphtha lene
55	Compound No.	37	38	36	40	41	42	\$	4	45

5	Position		٠ ا	I	I	I	1	l	I	1
	2	Ŧ	x	Ŧ	×	<b>±</b>	æ	æ	x	±
10	U	СООМе	H000	СООМе	H000	COOMe	H000	H000	H000	Н000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	8	8	8	8	8	8	8	8	8
25	>	NO2	N0 <sub>2</sub>	Ŧ	I	NO <sub>2</sub>	NO <sub>2</sub>	x	<b>±</b>	=
	∢	CH (OMe)	CH (OMe)	CH (OEt)	CH (OEt)	CH (OEt)	CH (OEt)	9	SO SO	805
30	R5	l l	ı	1	1	1	1	1	ı	
35	R4	1	1	1	ı	1	. 1	I	1	1
40	Position	l l	ı	ı	ı	ι	ı	ı	ι	ı
45	R1 and R2	ᆂ	Ŧ	Ŧ	Ŧ	Ŧ	±	Ŧ	Ŧ	+
50	×	Naphthalene	Naphthalene	Naphtha l ene	Naphtha l ene	Naphthalene	Naphtha!ene	Naphtha l ene	Naphtha!ene	Naphtha lene
55	Compound No.	46	47	48	49	20	51	52	53	54

	Position	l.	ĺ	1 .	Į	1	1	ı	1	
5	Pos									
	2	×	<b>=</b>	I	I	×	I	I	×	=
10	ឲ	H000	COOMe	COOEt	n8₊000 ·	C00H	x	Ю	SO <sub>2</sub> NH <sub>2</sub>	इ
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	8	03 <sup>2</sup> H3	CH2CO	03 <sup>2</sup> ₩	CH2CO	CH2CO	CH2CO	CH2C0	CH <sup>2</sup> CO
25	<b>+</b>	CN	x	<b>=</b>	æ	<b>=</b>	I	æ	=	<b>=</b>
	∢	CH <sub>2</sub> 0	0	0	0	0	0	0	0	0
30	R5	I	I	i	1	ŧ	1	1	I	1
35	<b>R4</b>	ı	I	1	1	1	t	1	1	I
40	Position	l	1	i	t	1	1 .	ι	1	1
	R1 and R2	Ŧ	Ŧ	Ŧ	Ŧ	Ŧ	干	Ŧ	Ŧ	Ŧ
45	R			<b>a</b> )	•	•		0)		
50	×	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphtha lene	Naphthalene	Naphtha lene	Naphthalene
55	Compound No.	55	99	23	28	29	09		62	83

5	Position	I	1	1	I	I	I	I	I	1
	7	æ	æ	Ŧ	æ	I	I	=	Ŧ	포
10	g	tet	000Me	H000	C00Me	COOEt	H000	СООЖе	H000	СООМе
15	E-Substitution	4-position	3-position	3-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	CH <sub>2</sub> CO	CH <sub>2</sub> CO	GH,20	CH,CO	GH,23	64,00	CH,CO	GH,CO	CH <sub>2</sub> CO
25	>-	==	<b>=</b>	æ	NO <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>	NH <sub>2</sub>	NH <sub>2</sub>	СООМе
	∢	0	0	0	0	0	0	0	0	0
30	85	. 1	1	. 1		1	ı	i	1	1
35	R4	 	ı	t	ı	1	1	. 1	1	
55	Position	l	1	1	1	. 1	1	ı	1	1
40	R1 and R2	±	土	<del>上</del>	±	±	Ŧ	Ŧ	Ŧ	£
45	<u>F</u>		_	_		_	-		_	
50	×	Naphtha i ene	Naphthalene	Naphtha   ene	Naphtha l ene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene
55	Compound No.	64	99	99	<i>L</i> 9	89	69	0/	71	. 72

Compound No.   X   R1 and R2   Position   R4   R5   A   Y   E   E-Substitution   G   Z   Position	55		50	45	40	35	30		25	20	15	10		5
Naphthalene         H         -         -         -         0         C00H         GH <sub>2</sub> CO         4-position         C00Me         H           Naphthalene         H         -         -         -         0         H         GH <sub>2</sub> CO         4-position         C00Me         NO <sub>2</sub> Naphthalene         H         -         -         -         0         H         GH <sub>2</sub> CO         4-position         C00Me         F           Naphthalene         H         -         -         -         0         H         GH <sub>2</sub> CO         4-position         C00Me         F           Naphthalene         H         -         -         -         0         H         GH <sub>2</sub> CO         4-position         C00Me         F           Naphthalene         H         -         -         -         0         H         GH <sub>2</sub> CO         4-position         C00Me         F	Compound	d No.	×	R1 and R2	Position	R4	R5	<	>	Ш	E-Substitution		7	Position
Naphthalene         H         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         NO <sub>2</sub> Naphthalene         H+         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         F           Naphthalene         H+         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         F           Naphthalene         H+         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         F           Naphthalene         H+         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         F           Naphthalene         H+         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         CI	73		Vaphthalene	Ŧ	I	ı	1	0	НООЭ	CH <sub>2</sub> CO	4-position	H000	Ŧ	
Naphthalene         H-         -         -         -         0         H         GH <sub>2</sub> CO         4-position         GOOMe         F           Naphthalene         H-         -         -         -         0         H         GH <sub>2</sub> CO         4-position         GOOMe         F           Naphthalene         H-         -         -         -         0         H         GH <sub>2</sub> CO         4-position         GOOMe         F           Naphthalene         H-         -         -         -         0         H         GH <sub>2</sub> CO         4-position         GOOMe         F           Naphthalene         H-         -         -         -         0         H         GH <sub>2</sub> CO         4-position         GOOMe         CI	74		Vaphthalene	Ŧ	ı	ı	1	0	=	CH2CO	4-position	СООМе	NO2	4-position
Naphthalene         H-         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         F           Naphthalene         H-         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         F           Naphthalene         H-         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         F           Naphthalene         H-         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         CI           Naphthalene         H-         -         -         -         0         H         CH <sub>2</sub> CO         4-position         COOMe         CI	75		Vaphthalene	Ŧ	ı	ı	1	<b>o</b> .	I	CH2CO	4-position	H000	NO <sub>2</sub>	4-position
Naphthalene         H-         -         -         0         H         CH2CO         4-position         COOMe         F           Naphthalene         H-         -         -         0         H         CH2CO         4-position         COOMe         F           Naphthalene         H-         -         -         0         H         CH2CO         4-position         COOMe         F           Naphthalene         H-         -         -         0         H         CH2CO         4-position         COOMe         Cl	76		Naphtha lene	Ŧ	l	ŧ	Į	0	x	CH <sub>2</sub> CO	4-position	COOMe	<b>L</b>	4-position
Naphthalene         H-         -         -         0         H         GH <sub>2</sub> CO         4-position         COOMe         F           Naphthalene         H-         -         -         0         H         GH <sub>2</sub> CO         4-position         COOMe         Cl           Naphthalene         H-         -         -         0         H         GH <sub>2</sub> CO         4-position         COOMe         Cl	11		Naphtha lene	Ŧ	ı	1	1	0	I	CH5CO	4-position	H000	<b>LL</b>	4-position
Naphthalene         H-         -         -         0         H         GH2C0         4-position         COOMe         Cl           Naphthalene         H-         -         -         0         H         GH2C0         4-position         COOMe         Cl           Naphthalene         H-         -         -         0         H         GH2C0         4-position         COOM         Cl	78		Vaphthalene	#	1	1	1	0	=	CH <sup>5</sup> CO	4-position	СООМе	<b>L</b> .	5-position
Naphthalene H 0 H GH2CO 4-position COOMe Cl	79		Naphtha l ene	±	I	ı	1	0	I	CH2CO	4-position	H000	LL.	5-position
Naphthalene H 0 H CH2CO 4-position COOH CI	80		Naphtha lene	Ŧ	í	1	l	0	x	CH2CO	4-position	COOMe	5	5-position
	18		Naphthalene	±	ì		ı	0	=	8 <sup>2</sup> 50	4-position	H000	5	5-position

5	Position	6-position	6-position	I	ı		ı	I	ŀ	-
	7	We was	Me	æ	Ŧ	Ŧ	<b>=</b>	<b>=</b>	I	=
10	IJ	СООМе	H000	СООМе	H000	C	tet	COOMe	H000	COOMe
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	CH <sub>2</sub> CO	CH2CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH2C0	CH2CO	CH <sub>2</sub> CO
25	>	×	<b>=</b>	<b>=</b>	<b>=</b>	±	æ	NO <sub>2</sub>	NO <sub>2</sub>	<b>±</b>
	. 4	0	0	S	S	S	S	S	S	₹
30	R5	I	ı	1	1	1	1	1	1	1
35	R4	I	I	1	i	1	- 1	1	1	1
	Position	ı	f	ı	ı	1	1	i	1	
40	Ri and R2	Ŧ	±	Ŧ	∓	±	土	Ŧ	干	±
45	RI									
50	×	Naphthalene	Naphthalene	Naphtha lene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene
55	Compound No.	82	83	84	85	98	87	88	88	06

5	Position		ı	I	1	I	I	I	ı	1	
	7	=	x	Ŧ	Ŧ	x	Ŧ	æ	Ŧ	±	
10	U	COOEt	H000	I	뜡	SO <sub>2</sub> NH <sub>2</sub>	tet	СООМе	H000	СООМе	
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
20	Ш	CH <sub>2</sub> CO	CH <sub>2</sub> C0	CH <sup>2</sup> CO	CH2C0	CH2CO	CH2CO	CH5CO	84,50	CH,CO	
25	<b>&gt;</b>	盂	<b>=</b>	±	Ŧ	Ŧ	æ	NO <sub>2</sub>	NO <sub>2</sub>	<b>=</b>	
	<	£,	₹	중	₹	ਝੌ	ર્સ	ਲੌ	ਠੌ	0CH <sub>2</sub>	
30	R5	ı	1		1	1	1	1	1	ı	
35	<b>84</b>	 	1	·	I	1	ł	I	. 1	ı	
	Position	ı	1	I	l	l	ı	1	1	ı	
40			ı	,	ı	1	1	1	1		
45	R1 and R2	ᆂ	Ŧ	Ŧ	Ŧ	포	于	Ŧ	Ŧ	Ŧ	
50	×	Naphtha lene	Naphtha lene	Naphtha lene	Naphtha   ene	Naphthalene	Naphtha i ene	Naphtha lene	Naphtha i ene	Naphtha l ene	
55	Compound No.	16	92	93	94	95	96	97	86	66	

5	Position	1	l	I	l	l	ŀ	I	I	ı
	2	±	<b>=</b>	<b>=</b>	x	æ	±	I	<b>=</b>	=
10	IJ	H000	СООМе	H000	СООМе	H009	COOMe	H000	СООМе	H000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	CH,CO	CH2CO	CH2CO	CH <sub>2</sub> CO	CH2CO	CH <sub>2</sub> CO	CH2C0	CH <sup>2</sup> CO	CH2C0
25	<b>&gt;</b>	æ	±	<b></b>	±	æ	Ξ	±	NO <sub>2</sub>	NO <sub>2</sub>
	∢	00H <sub>2</sub>	SGH2	SGH <sub>2</sub>	8	8	CH (OMe)	CH (OMe)	CH (OMe)	CH (OMe)
30	R5	l	1	1	1	ŀ	1	1	1	i
35	R4	I	. 1	i	•	1	ŧ	1	. 1	
40	2 Position	ı	i	·	i	i	1	1	I	1
45	R1 and R2	土	土	于	Ŧ	Ŧ	ᆍ	Ŧ	干	Ŧ
50	×	Naphthalene	Naphtha lene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene
55	Compound No.	100	101	102	103	104	105	106	107	108

						_				
5	Position	ı	. 1	1	I	5-position	1	1	l	1
	2	±	±	Ŧ	<b>=</b>	We .	Œ	I	I	=
10	g	сооме	H000	. СООМе	H000	H000	СООМе	H000	COOMe	Н000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	02 <sub>4</sub> 20	CH <sub>2</sub> C0	CH2C0	CH <sup>2</sup> CO	CH500	CH2CH2CO	CH2CH2CO	CH2CH2CO	CH2CH2CO
	>	×	=	NO <sub>2</sub>	NO <sub>2</sub>	<b>=</b> .	×	Ŧ	=	=
25	4	CH (0Et)	CH (OEt)	CH (OEt)	CH (OEt)	0	0	0	S	S
30	R5	I	1	1	1	t	1	1	•	
35	R4	1	1	. 1	ı	i	1	1	1	1
40	Position	l	ı	1	I	I	l		1	1
45	R1 and R2	Ŧ	£	±	±	Ŧ	±	Ŧ	Ŧ	+
50	×	Naphthalene	Naphthalene	Naphthalene	Naphtha!ene	Naphtha lene	Naphthalene	Naphtha l ene	Naphthalene	Naphtha lene
55	Compound No.	109	110	Ξ	112	113	114	115	116	117

5	Position	1	1	1	I	1	1	ł	i	ı
	И	Ŧ	æ	Ŧ	×	x	=	×	x	=
10	g	СООМе	C00H	COOMe	H000	СООМе	H000	СООМе	H000	СООМе
15	E-Substitution	4-position								
20	ш.	CH2CH2CO	CH2CH2CO	CHECHCO	CH=CHC0	CHECHCO	CH=CHCO	CH=CHC0	CH=CHC0	CH (Me) CO
25	>	工	±	¥	æ	=	<b>=</b>	<b>=</b>	=	=
	<	ર્કે	<u>ਝ</u>	0	0	S	S	₹	£	0
30	. R5	ı	I	I	. 1	1	I	1	1	. 1
35	25	1	J	l	. 1	1	İ	ı	. 1	1
40	Position	1	1	1	1	1	I	1	I	1
45	R1 and R2	土	Ŧ	Ŧ	Ŧ	Ŧ	土	Ŧ	· 士	±
50	×	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphtha ene	Naphthalene	Naphthalene	Naphthalene
55	Compound No.	118	119	120	121	122	123	124	125	126

5	Position	ı	ţ	ı	I	l	1	t	l	1
	2	=	I	I	×	<b>x</b>	I	I	I	=
10	ŋ	НООО	сооме	H000	H000	H000	000Me	H000	COOMe	Н000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	CH (Me) CO	C (Me) 2C0	C (Me) 2C0	CH5C0	CH2C0	8	8	8	00
25	>	æ	. <b>=</b>	Ŧ	I	I	I	Ŧ	NO <sub>2</sub>	NO <sub>2</sub>
	∢	0	0	0	0	0	0	0	0	0
30	R5	I	;	I	I	I	I	1	1	ı
35	R4	l	1	i	1	I	I	1	1	ı
	Position	1	1	ı	6-position	6-position	6-position	6-position	6-position	6-position
40						ဖ်	, ,	φ.	Ġ	
45	R1 and R2	土	±	土						
50	- ×	Naphthalene	Naphthalene	Naphthalene	Naphthalene Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene
55	Compound No.	127	128	129	130	131	132	133	134	135

5	Position	l	I	<b>I</b>	I .	1	ı	ı	1	ı
	7	Ŧ	I	I	I	x	x	<b>=</b>	I	=
10	Œ	СООМе	H000	СООМе	Н000	СООМе	H000	сооме	Н000	СООМе
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH2CO	8	8	8	8	CH2C0
25	>	王	I	NO <sub>2</sub>	NO <sub>2</sub>	Œ	Ξ	NO <sub>2</sub>	NO <sub>2</sub>	æ
	∢	0	0	0	0	0	0	0	0	0
30	R5	l	I	I	I	1	ı	1	T <sub>e</sub>	1
35	on R4	lon	l I	ion	l I	l noi	l noi	l noi	ion	lon
40	2 Position	6-position	6-position	6-position	6-position	∼o∕ 6-position	∼o∕ 6-position	^o∕ 6-position	∼o∕ 6-position	o 6-position
45	R1 and R2						<b>→</b>		~	
50	×	Naphtha   ene	Naphthalene	Naphthalene	Naphtha lene	Naphtha ene	Naphthalene	Naphtha ene	Naphthalene	Naphtha I ene
55	Compound No.	136	137	138	139	140	141	142	143	144

	Position	1	í	ł	1	ı	ı	1	ţ	,
5	Z Pos	<b>.</b>	×	Ŧ	æ	I	I	<b></b>	æ	_
	"	_		_		_	_			
10	g	COOH	COOMe	H000	СООМе	C00H	СООМе	H000	H000	Н000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	CH2CO	CH2CO	CH <sub>2</sub> CO	8	8	02 <sup>7</sup> H5	CH2CO	8	CH <sub>2</sub> CO
25	>	æ	NO <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>	æ	æ
	4	. 0	0	0	0	0	0	0	0	0
30	R5	1	t	ı	ı	1	1	1	1	ı
	R4	1	1	1	1	1	1	ı	ì	1
35		tion	tion	tion	tion	tion .	tion	tion	tion	tion
	Position	6-position	o 6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position
40	2	<b>)</b>	۶	۶	Ъ	<b>`</b>	<b>\</b>	þ		
	R1 and R2		$\rightarrow$	$\stackrel{>}{\sim}$	<u></u>				$\stackrel{>}{\sim}$	3
45		- <	<del>-</del> <	<del>-</del> (	<u>'</u>	<b></b> ./	<b>_</b>	• • • • • • • • • • • • • • • • • • •	w	
50	×	Naphtha l ene	Naphtha!ene	Naphthalene	Naphthalene	Naphtha lene	Naphthalene	Naphtha lene	Naphtha lene	Naphthalene
55	Compound No.	145	146	147	148	149	150	151	152	153

	Position	1	1	1	ı	ı	ŀ	1	1	ı
5										
	7	=	. =	×	Ξ	=	I	=	I	. <b>=</b>
10	O	Н000	СООМе	H000	СООМе	H000	COOMe	H000	СООМе	H005
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	CH2CO	8	8	9	8	CH <sub>2</sub> CO	CH2CO	CH <sub>2</sub> CO	CH2CO
25	>	Ŧ	x	æ	NO <sub>2</sub>	NO <sub>2</sub>	<b>±</b>	I	N0 <sub>2</sub>	NO <sub>2</sub>
	∢	0	0	0	0	0	0	0	0	0
30	85	I	1	I	ł	I	1	1	. 1	1
<i>35</i>	R4	1	1	i	ı	1	1	i	ŀ	ı
	Position	7-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position
40	R1 and R2	~	<b>&gt;</b>	<b>&gt;</b>	\ \ \	\ \ \		\ \ \	\ \ \	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \
45	R	/								
50	×	Naphtha lene	Naphthalene	Naphtha lene	Naphthalene	Naphtha lene	Naphtha lene	Naphtha lene	Naphthalene	Naphtha lene
55	Compound No.	154	155	156	157	158	159	160	161	162

	Position	1	1	1	1	,	1	ì	ì	1
5	2 P	=	<b>=</b>	<b>=</b>	<b>=</b>	<del>=</del>	<b>=</b>	==	=	±
10	ত	COOMe	H000	СООМе	H000	COOMe	C00H	СООМе	Н000	сооме
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	8	8	8	8	CH,200	CH <sub>2</sub> CO	CH2C0	CH,200	8
25	>	æ	æ	NO <sub>2</sub>	NO <sub>2</sub>	Ŧ	<b>±</b>	NO <sub>2</sub>	NO <sub>2</sub>	æ
	∢	0	. 0	0	0	0	0	0	0	0
30	£2	1	1	I	ı	ı	ı	i	I	1
35	R4	1	1	1	ı	I	I	i	1	1
	Position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position
40		\ \rangle \r	<b>,</b>	> >	> •	> >	> >	> >	> >	<b>,</b>
45	R1 and R2									
50	<b>×</b>	Naphthalene	Naphtha ene	Naphthalene	Naphtha lene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene
55	Compound No.	163	164	165	166	167	168	169	170	11.1

5	Position	1	1	1	1	1	1	1	i	1
	2	=	<b>x</b>	Ŧ	Ξ	I	<b>=</b>	I	<b>=</b>	Ŧ
10	g	HOOO	СООМе	H000	COOMe	Н000	COOMe	Н000	H000	COOMe
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	8	8	8	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH2CO	CH <sub>2</sub> C0	8
25	>	æ	NO <sub>2</sub>	NO <sub>2</sub>	x	Ŧ	NO <sub>2</sub>	NO <sub>2</sub>	<b>=</b>	Ŧ
	<	0	0	0	0	0	0	0	0	0
30	R5	l	1	I	ŀ	1	1	1	1	- 1
35	R4	l E	i E	, I	ا ج	l E	5	I	l E	- uo
40	Position	6-position	° 6-position	° 6-position	° 6-position	° 6-position	6-position	o′6-position	o'6-position	o 6-position
45	R1 and R2								\{\}	-},
50	×	Naphthalene (	Naphthalene (	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphtha!ene	Naphthalene
55	Compound No.	172	173	174	175	9/1	171	178	179	180

	1									1
5	Position	1	ł	1	l	l	1	1	l	1
	2	±	=	æ	<b>=</b>	<b>=</b>	Ŧ	I	=	=
10	ŋ	H000	COOMe	H000	COOMe	H000	COOMe	H000	COOMe	H000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	8	8	8	CH2CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	8	8
25	>	Ŧ	NO <sub>2</sub>	NO <sub>2</sub>	Ξ	<b>=</b>	NO <sub>2</sub>	NO <sub>2</sub>	Ŧ	Ŧ
	∢	0	0	0	0	0	0	0	0	0
30	R5	I	1	1	ı	ı	1	í	1	ı
	R4	ŀ	1	I	1	1	1	ı	Į.	
35	Position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position
40	P <sub>Q</sub>	9	Ţ	9	9	Ţ	J	Ţ	<b>G</b>	Ţ.
45	R1 and R2	$\Rightarrow$	$\stackrel{>}{\prec}$	$\stackrel{>}{\sim}$	$\prec$	$\stackrel{>}{\prec}$	$\stackrel{>}{\prec}$	$\stackrel{>}{\prec}$		>
50	×	Naphtha lene	Naphtha lene	Naphtha lene	Naphtha lene	Naphtha!ene	Naphthalene	Naphtha l ene	Naphthalene	Naphtha l ene
<i>55</i>	Compound No.	181	182	183	184	185	186	187	188	189

5	Position	ı	1	I	I	I	1	I	I	ı
	7	I	Ŧ	I	I	I	I	<b>=</b>	I	=
10	g	COOMe	H000	СООМе	H000	СООМе	H000	H000	H000	C00H
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	8	8	CH2CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH2CO	8	CH2CO
25	<b>&gt;</b>	NO <sub>2</sub>	NO <sub>2</sub>	<b>=</b>	<b>±</b>	NO <sub>2</sub>	NO <sub>2</sub>	×	æ	=
	4	0	0	0	0	0	0	0	0	0
30	RS	ı	1	1	1	1	1	l	ı	
35	R4	l	1	1	1	1 .	1	1	1	1
40	Position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position
45	R1 and R2	>		<b>&gt;</b>			\ \ \	~ ·	Ò	3
50	×	Naphtha lene	Naphtha l ene	Naphtha lene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene
55	Compound No.	190	191	192	193	194	195	196	197	198

5	Position	1	I	1	l .	I	I	1	I	1
	Z	Ŧ	Ŧ	I	I	<b>=</b>	I	±	±	=
10	5	H000	H000	СООЖе	C00H	СООМе	H000	емооэ	H000	СООМе
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	02 <sup>7</sup> K9	CH <sup>2</sup> CO	8	8	8	8	CH <sub>2</sub> CO	CH2C0	G <sup>2</sup> CO
25	<b>&gt;</b>	×	±	×	×	NO <sub>2</sub>	NO <sub>2</sub>	æ	<b>=</b>	NO <sub>2</sub>
	<	0	0	0	0	0	0	0	0	0
30	85	l	1	1	1	f	1 .	1	ı	J
0.5	25	ı	ŧ	I	I	I	1	F,	1	,
35	Position	7-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position
40	<b>K</b> 2		,					,	<b>\</b>	
45	R1 and R2	ď	<b>&gt;</b>	ं	ð	ं	ð	ै	ð	ď
50	×	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphtha ene	Naphthalene	Naphthalene	Naphtha lene	Naphthalene
55	Compound No.	199	200	201	202	203	204	205	506	207

	Position		ı	1	1	1	1	ł	ı	
5	Posi		'	•	•	•	•	·	·	
	7	=	æ	Ŧ	=	æ	Œ	<b>=</b>	×	=
10	g	H000	H000	H000	H000	C00H	COOMe	C00H	COOMe	H000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	CH <sub>2</sub> CO	CH <sub>2</sub> CO	8	CH <sub>2</sub> CO	05 <sub>4</sub> 0	8	8	8	8
25	>	NO <sub>2</sub>	æ	Ŧ	×	Ŧ	<b>=</b>	=	NO <sub>2</sub>	NO <sub>2</sub>
	∢	0	0	0	0	0	0	0	0	0
30	R5	,	1	1	ı	I	ı	I	ı	ı
35	R4	1	i	1	1	l	i	ı	l	1
40	Position	6-position	7-position	6-position	6-posítion	6-position	6-position	6-position	6-position	6-position
45	R1 and R2	   	ें		ð	Ò	웊	. 호.	- <del>0</del>	-0H
50	×	Naphtha l ene	Naphtha l ene	Naphthalene	Naphtha lene	Naphtha   ene	Naphtha lene	Naphtha I ene	Naphtha lene	Naphtha l ene
55	Compound No.	208	209	210	211	212	213	214	215	216

5	Position	1	1	١	1	ı	i	1	1	1	
	7	I	I	Ŧ	I	<b>±</b>	<b>±</b>	<b>=</b>	¥	<b>=</b>	
10	ប	COOMe	H000	СООМе	C00H	COOMe	C00H	емооэ	H000	СООМе	
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
20	Ш	CH2CO	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH,CO	8	8	CH2CO	CH2CO	8	
25	>	±	×	NO <sub>2</sub>	NO <sub>2</sub>	Ŧ	æ	æ	±	エ	
	∢	0	0	0	0	0	0	0	0	0	
30	R5	I	ı	1	1	I	1	1	ı	ı	İ
35	R4	1	1	1	1	ŀ	1	1	1	i	
40	Position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	
45	R1 and R2	-6 <del>.</del>	H9-	-0 <del>2</del>	호						
50	×	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphtha lene	Naphthalene	Naphthalene	Naphtha I ene	
55	Compound No.	217	218	219	220	221	222	223	224	225	

5	Position	1	1	I	I	1	I	I	1	ı
	2	=	Ŧ	x	=	×	Ŧ	<b>=</b> .	<b>=</b>	=
10	Ø	H000	COOMe	H000	СООМе	H000	COOMe	Н000	СООМе	Н000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	05	8	8	8	8	8	8	CH <sup>2</sup> CO	CH <sub>2</sub> CO
25	<b>&gt;</b>	=	Ŧ	±	NO <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>	NO <sub>2</sub>	<b>=</b>	×
	∢	0	0	0	0	0	0	0	0	0
30	R5	1	I	I	t	1	i	1	1	1
<i>35</i>	<b>R4</b>	ı	ŧ	1	1	ı	1	. 1.	ı	l
40	Position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position	6-position
45	R1 and R2									
50	×	Naphthalene	Naphthalene	Naphtha lene	Naphthalene	Naphtha lene	Naphthalene	Naphthalene	Naphthalene	Naphthalene
55	Compound No.	.226	227	228	229	230	231	232	233	234

5	Position	ı	1	I	Ī	I	1	ı	1	1
	N	포	x	I	Ŧ	×	<b>±</b>	<b>=</b>	I	z
10	5	СООМе	H000	H000	СООМе	C00H	COOMe	H000	СООМе	Н000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	CH <sub>2</sub> CO	CH <sub>2</sub> C0	CH2CO	8	8	8	8	CH200	CH2C0
25	<b>&gt;</b>	NO <sub>2</sub>	NO <sub>2</sub>	<b>±</b>	Ŧ	=	NO <sub>2</sub>	NO <sub>2</sub>	×	=
	<	0	0	0	0	0	0	0	0	0
30	R5	1		1	ı	I	t	1 .	1	ı
35	R4	1	1	ı	l	I	1	ı	. 1	l
40	Position	6-position	6-position	7-position	6-position	6-position	6-position	6-position	6-position	6-position
45	R1 and R2				CH <sub>3</sub> O-	CH <sub>3</sub> 0-	CH <sub>3</sub> O-	<u>유</u>	CH <sub>3</sub> O-	CH <sub>3</sub> O-
50	×	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene	Naphthalene
<i>55</i>	Compound No.	235	236	237	238	239	240	241	242	243

5	Position	I	I	1	I	I	I	I	1	ı
	7	Ŧ	Ŧ	<b>. =</b>	Ŧ	Ŧ	Ξ	Ŧ	x	=
10	g	COOMe	H000	C00H	H000	H000	H000	H000	C00H	СООЖе
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	CH5CO	CH2CO	8	8	8	8	CH <sub>2</sub> CO	CH2C0	CH2CO
25	>	NO <sub>2</sub>	NO2	<b>±</b>	=	<b>=</b>	æ	Ξ	×	<b>=</b>
20	∢	0	0	8	CH (OMe)	ŧ	<b>0</b>	0	0	0
30	85	1	1	×	I	Ŧ	<b>±</b>	æ	æ	=
<i>35</i>	R4	I	1	±	<b>=</b>	±	x	<b>=</b>	æ	=
	Position	6-position	6-position	ı	1	I	t	I	4-position	4-position
40	R1 and R2	СН <sub>3</sub> 0-	G,90	×	×	Œ	×	<b>=</b>		<u>}</u>
50	×	Naphthalene	Naphthalene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene
55	Compound No.	244	245	246	247	248	249	250	251	252

	Position		1	1	1	1	1	1	1	, 1
5	Pos		·			·	·	•	•	·
	N	=	<b>=</b> .	=	I	×	. <b>=</b>	Ŧ	==	=
10	g	H000	H000	COOMe	H000	СООМе	H000	СООМе	H000	COOMe
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	CH2CO	CH2C0	CH2CO	CH,CO	CH <sub>2</sub> CO	CH2CO	CH <sub>2</sub> CO	<b>G</b> <sup>2</sup> CO	CH <sub>2</sub> CO
25	<b>&gt;</b>	玉	x	Ŧ	æ	I	. <b>=</b>	×	I	=
	∢	0	0	0	0	0	0	0	0	0
30	R5	æ	I	æ	x	æ	=	Ŧ	Ŧ	<b>=</b>
35	. R4	Ŧ	I	<b>=</b>	æ	æ	×	×	I	=
	Position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
40	22		<b>,</b>			þ	þ			
45	R1 and R2	<u>}</u>		<b>\</b>	<b>\( \)</b>			<u>}</u>	<u>}</u>	
50	×	Вепzепе	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene
55	Compound No.	253	254	255	256	257	258	259	260	261

	6	l							•	
5	Position	1	1	l	I	Î	l	t·	1	ı
	N	Ŧ	×	æ	I	I	æ	æ	<b>=</b>	=
10	g	C00H	Н000	H000	СООМе	Н000	H000	C00H	COOMe	Н000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Е	CH <sub>2</sub> CO	8	8	CH <sub>2</sub> CO	CH <sub>2</sub> CO	CH2C0	8	CH2C0	CH <sup>2</sup> CO
25	>	æ	ક	×	<b>=</b>	×	<b>x</b>	<b>=</b>	æ	±
	4	0	0	0	0	0	0	0	0	0
30	R5	Ŧ	x	æ	<b>=</b>	×	±	<b>=</b> '	±	Ŧ
<i>35</i>	R4	王	×	=	<b>=</b>	I	<b>=</b>	×	×	±
	Position	4-position	4-position	4-position	4-position	4-posítion	4-position	4-position,	4-position	4-position
40		4	. 4	4	4	4	4	þ	4	4
45	R1 and R2	\ \ \	3	$\prec$	7	7	~	}	\ \ \	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
50	×	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene
55	Compound No.	262	263	264	265	266	267	268	569	270

5	Position	1	į	1	I	1	ı	I	I	1	
	Ν	<b>#</b>	æ	I	Ŧ	Ŧ	±	I	×	=	
10	ر ت	Н000	СООМе	6 Н000	COOMe	H000	СООМе	H000	COOMe	H000	
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
20	ш	8	CH <sub>2</sub> CO	CH <sub>2</sub> CO	8	8	8	8	CH2CO	6 <del>4</del> ,60	
25	<b>&gt;</b>	#	I	x	æ	<b>=</b>	NO <sub>2</sub>	N0 <sub>2</sub>	æ	<b>=</b>	
	∢	0	0	0	0	0		0	0	0	
30	R5	Ŧ	Ŧ	I	Ŧ	<b>=</b>	I	æ.	<b>=</b>	=	
35	R4	=	<b>エ</b> ・	I	Ŧ	×	I	Ŧ	<b>=</b>	×	
40	Position	4-position	4-posítion	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
45	R1 and R2	<b>&gt;</b>	~	~	ð	$\overrightarrow{\bigcirc}$	o o	ð	ð	ð	
50	×	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	
55	Compound No.	271	272	273	274	275	276	277	278	279	

5	Position	5-position	5-position	5-position	1	1	1	ı	ĭ	ı
	2	. 15	E e	Me .	<b>±</b>	x	I	±	I	=
10	Ŋ	H000	СООМе	H000	СООМе	H000	сооже	H000	СООМе	СООН
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	ш	CH2CO	CH <sub>2</sub> CO	<b>6</b> 4500	CH,CO	CH200	CH2C0	ଫ୍ଲେପ	CH <sub>2</sub> CO	CH <sub>2</sub> CO
25	<b>&gt;</b>	Ŧ	<b>=</b>	=	=	<b>±</b>	æ	<b>±</b>	±.	±
	4	0	0	0	0	0	0	0	0	0
30	R5	Ŧ	<b>=</b>	×	×	×	æ	Ŧ	æ	<b>=</b>
35	R4	=	=	Ξ	Ξ	I	I	I	<b>=</b>	=
	Position	4-position	4-position	4-position	2-position	2-position	4-position	4-position	4-position	4-position
40	R1 and R2			Š	Š	Š		}	ð	강
45	\ \&		$\smile$	$\bigcirc$			~	-		
50	×	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene
55	Compound No.	280	281	282	283	284	285	286	287	288

5	Position	1	1	I	I	1	i	i	1	ı	
	2	=	Ŧ	x	<b>=</b>	<b>=</b>	<b>=</b>	Ŧ	×	=	
10	5	НООО	H000	COOMe	C00H	H000	COOMe	H000	COOMe	H000	
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
20	ш	CH2CO	CH2CO	CH2CO	CH2CO	CH200	8	8	ଫ୍ଲେଷ	CH <sub>2</sub> CO	
25	>	Ŧ	Ŧ	×	x	Ŧ	Ŧ	×	Ŧ	æ	
	. ∢	0	0	0	0	0	0	0	0	0	
30	R5	π.	Ŧ	x	æ	Œ	<b>=</b>	I	I	±	
	R4	ェ	Ŧ	<b>=</b>	æ	×	<b>±</b>	æ	=	±	
35	Position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
40		4	7		4	4	•	4	4	4	
45	R1 and R2	6	ð	ð	ð	CF <sub>3</sub> -	9	훋	훋	오	
50	×	Benzene	Benzene	Вепzепе	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	
55	Compound No.	289	290	291	292	293	294	. 295	296	297	

5	Position	1	f	I	ı	I	I	1	1	I	
	2	Ξ	<b>=</b>	<b>=</b>	x	x	I	=	Ŧ	<b>=</b>	
10	IJ	H000	H000	H000	H000	H000	H000	C00H	H000	C00H	
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
20	Ш	8	8	8	8	8	8	8	8	8.	
25	<b>&gt;</b>	×	x	5	C	NHCOCH <sub>3</sub>	NHS0 <sub>2</sub> CH <sub>3</sub>	ğ	æ	Ŧ	
	∢	0	0	0	0	0	0	0	0	0	
30	RS	<b>=</b>	×	==	Ŧ	æ	<b>±</b>	Ŧ	×	10	
	R4	=	=	I	Ŧ	×	±	=	5	Ξ	
35	Position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
40	R2	)   	ъ Ъ	þ	þ	þ		,	`	,	
45	R1 and R2										
50	×	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	
55	Compound No.	298	299	300	301	302	303	304	305	306	

5	Position	1	1	l	١	I	ı	1	1	1
	7	±	×	×	=	<b>=</b>	<b>x</b>	×	Ŧ	±
10	5	Н000	H000	H000	H000	H000	H000	сооже	H000	COOMe
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	8	8	8	8	8	8	CH2CO	CH2C0	СН <sup>2</sup> СО
25	<b>&gt;</b>	<b>=</b>	×	I	×	±	NO <sub>2</sub>	±	<del>=</del>	=
:	∢	0	0	0	0	0	0	S	S	0
30	R5	Ŧ	0CH3	<b>=</b>	CH3C0	కో	I	<b>±</b>	±	<b>=</b>
	<b>R4</b>	9643	±	S. F.	×	ૠ૾ૼ	<b>=</b>	<b>=</b>	Ŧ	=
35	Position	4-position	4-position	4-position	4-position	4-position	4-posítion	4-position	4-position	4-position
40	22				,	\.	_	\_ \	_	
45	R1 and R2									
50	×	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene
55	Compound No.	307	308	309	310	311	312	313	314	315

	Position		5-position	5-position	4-position	4-position	5-position	5-position	ı	,	
5	Pos		5-po	5-po	4-po	4-po	5-po	5-po			
	N	=	5	5	LL.	L	₩e	We was	×	æ	
10	g	H000	СООМе	H000	СООЖе	H000	СООМе	H000	СООМе	C00H	
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position	
20	ш	CH <sub>2</sub> CO	CH2C0	CH <sub>2</sub> CO	CH2CO	CH2CO	CH2CO	CH <sub>2</sub> CO	CH2C0	CH2CO	
25	>	=	æ	=	<b>=</b>	×	<b>=</b> ,	<b>=</b>	æ	. <del>=</del>	
	∢	0	0	0	0	0	0	0	0	0	
30	RS	=	=	Ŧ	æ	×	Ŧ	±	<b>±</b>	=	
	₩.	=	=	<b>=</b>	<b>=</b>	=	=	<b>3</b> =	<b>=</b>	=	
35	Position	4-position	3-position	3-position	3-position	3-position	3-position	3-position	2-position	2-position	
40			, m			ris .	, ri	ю	4	. 2	
45	R1 and R2										
50	<b>×</b>	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	
<i>55</i>	Сопроипа No.	316	317	318	319	320	321	322	323	324	

	ي								
5	Position	1	1	1	I	1	1	1	-
	ν.	x	x	<b>=</b>	<b>=</b>	x	I	=	=
10	g	H000	H000	H000	H000	H003	H000	H000	Н000
15	E-Substitution	4-position	4-position	4-position	4-position	4-position	4-position	4-position	4-position
20	Ш	CH2CO	CH <sup>2</sup> CO	CH <sub>2</sub> CO	05 <sup>7</sup> H9	CH <sub>2</sub> CO	GH2CH2CO	CH=CHCO	CH <sup>2</sup> CO
25	>	=	x	Ŧ	×	Ŧ	<b>=</b>	· =	±
25	∢	0	S0 <sub>2</sub>	8	0	0	0	0	0
30	85	王	I	Ŧ	Ŧ	0CH <sub>3</sub>	, <b>=</b>	I	ı
	\$	æ	<b>=</b>	Ŧ	00H <sub>3</sub>	<b>=</b>	I	<b>=</b>	1
35	Position	3-position	4-position	4-position	4-position	4-position	4-position	4-position	6-position
40			4	4		4	4		(9)
45	R1 and R2								8
50	×	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Benzene	Naphtha l ene
55	Compound No.	325	326	327	328	329	330	331	332

[0035] The anthranilic acid derivative that is an active ingredient of the cancer remedy of the present invention has a strong cytotoxic activity as described in the Examples hereinafter. Specifically, the anthranilic acid derivative has an  $LC_{50}$  or a  $GI_{50}$  of 5 $\mu$ M or below, preferably 0.05 nM or above and 1 $\mu$ M or below, more preferably 0.05 nM or above and 500 nM or below.

[0036] The anthranilic acid derivative having the excellent cytotoxic activity can be used as an active ingredient of the remedy clinically applicable to cancer.

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[0037] Furthermore, the anthranilic acid derivative or a pharmaceutically acceptable salt thereof represented by the formula (1) can be produced if the persons are those skilled in the art by referring to WO95/32943 and WO97/19910. Namely, as shown in the following scheme, the objective compound represented by the following formula [I] can be obtained by condensing a carboxylic acid [II] having a naphthalene skeleton or a carboxylic acid [III] having a benzene skeleton with an aniline derivative [IV].

$$X \cdot A \downarrow \qquad \qquad X \cdot$$

[0038] R<sup>1</sup>, R<sup>2</sup>, X, A, Y, E, G and Z in each formula mentioned above are the same as defined above. E' represents a single bond or a bond; CR<sup>10</sup>R<sup>11-</sup>, -CH<sub>2</sub>CH<sub>2</sub>- or CH=CH-, wherein R<sup>10</sup> and R<sup>11</sup> are the same as defined above. The compounds, which are starting materials, represented by the formula [II] and formula [III] can be obtained according to a known method. The method for condensation represents roughly classified into a method for passing through an acid halide and a method without passing through the acid halide. Both the methods are basically known.

[0039] When the acid halide is passed, the compound [I] can be obtained by reacting the compound [II] or [III] with oxalyl chloride or thionyl chloride in the presence or absence of an additive such as DMF, producing the acid halide of the compound [II] or [III] and reacting the resultant acid halide with the compound [IV] in the presence or absence of a base.

[0040] On the other hand, in the method without passing through the acid halide, the compound [I] can be obtained by activating the compound [II] or [III] using various activators such as mixed acid anhydrides, carbodiimides, imidazole-forming agent, halophosphoric esters or cyanophosphoric esters and reacting the activated compound [II] or [III] with the compound [IV].

[0041] In the compound [I] thus obtained, when Y represents -COOR<sup>7</sup> and R<sup>7</sup> represents a lower alkyl group or G represents -COOR<sup>3</sup> and R<sup>3</sup> represents a lower alkyl group, the compound [I], if necessary, can be hydrolyzed under acidic or basic conditions and converted into a compound wherein R<sup>7</sup> or R<sup>3</sup> represents a hydrogen atom.

[0042] In the compound [I] thus obtained, when G represents -CN, the compound [I], if necessary, can be subjected to a treatment such as reaction with an azide compound and converted into a compound wherein G represents a tetrazol-5-yl group.

45 [0043] Furthermore, the compound [I] thus obtained (when Y represents -COOR<sup>7</sup> and R<sup>7</sup> represents a hydrogen atom or when G represents -COOR<sup>3</sup> and R<sup>3</sup> represents a hydrogen atom or G represents the tetrazol-5-yl group), if necessary, can be converted into the pharmaceutically acceptable salt described above.

[0044] Therefore, the anthtranilic acid derivative represented by the formula (1) or a pharmaceutically acceptable salt thereof which is an active ingredient of the cancer remedy of the present invention can be obtained.

[0045] The cancer remedy of the present invention can be administered orally or parenterally such as intravenously, subcutaneously, intramuscularly, percutaneously, intrarectally or by instillation or by inhalation.

[0046] Examples of the dosage form for oral administration include a tablet, a pill, a granule, a powder, a solution, a suspension, a syrup, a capsule and the like.

[0047] The tablet form can be produced according to a conventional method using, for example a vehicle such as lactose, starch or crystalline cellulose; a binder such as carboxymethyl cellulose, methyl cellulose or polyvinylpyrrolidone; a disintegrating agent such as sodium alginate, sodium hydrogencarbonate or sodium lauryl sulfate.

[0048] The pill, granule and powder can similarly be formed according to a conventional method using the vehicle and the like.

[0049] The solution, suspension and syrup can be formed according to a conventional method using glycerol esters, for example tricaprylin or triacetin; alcohols, for example ethanol; water; vegetable oils, for example corn oil, cottonseed oil, coconut oil, almond oil, peanut oil and olive oil.

[0050] The capsule is formed by filling a granule, a powder, a solution or the like in a capsule such as gelatin.

[0051] The dosage form for intravenous, subcutaneous or intramuscular administration includes a parenteral injection in the form of an aseptic aqueous or nonaqueous solution or the like. For example, an isotonic sodium chloride solution is used as the aqueous solution. For example, propylene glycol, polyethylene glycol, a vegetable oil such as olive oil and an injectable organic ester such as ethyl oleate are used as the nonaqueous solution. An isotonic agent, a preservative, a wetting agent, an emulsifying agent, a dispersing agent, a stabilizer and the like, if necessary, are added to the pharmaceutical preparation and the resulting pharmaceutical preparation can be sterilized by suitably carrying out treatment such as filtration through a bacterial filter, formulation of a disinfectant, heating, irradiation or the like. An aseptic solid pharmaceutical preparation is produced and can be used by dissolving the resulting pharmaceutical preparation in aseptic water or an aseptic solvent for injection just before use.

[0052] Examples of the dosage form for percutaneous administration include an ointment and a cream. The ointment is formed by using oils and fats such as castor oil and olive oil; vaseline and the like. The cream is formed according to a conventional method using a fatty oil; diethylene glycol; an emulsifying agent such as a sorbitan monofatty acid ester, and the like.

[0053] A usual suppository such as a gelatin soft capsule is used for rectal administration.

[0054] The dosage form of the eye drop includes an aqueous or a nonaqueous eye drop. Sterilized purified water, an isotonic sodium chloride solution or a suitable aqueous solvent is used as a solvent in the aqueous eye drop, and examples of the eye drop include an aqueous eye drop using only sterilized purified water as the solvent; a viscous eye drop prepared by adding a mucilage such as carboxymethyl cellulose, methyl cellulose, hydroxypropyl cellulose or polyvinylpyrrolidone; an aqueous suspension eye drop obtained by adding a suspending agent such as a surfactant or a polymeric thickener; a solubilized eye drop and the like prepared by adding a solubilizing agent such as a nonionic surfactant. The nonaqueous eye drop uses a nonaqueous solvent for injection as the solvent, and examples of the nonaqueous eye drop include a nonaqueous eye drop using a vegetable oil, a liquid paraffin, a mineral oil, proplylene glycol or the like; a nonaqueous suspension eye drop obtained by carrying out suspension using a thixotropic colloid such as aluminum monostearate and the like. An isotonic agent, a preservative, a buffer, an emulsifying agent, a stabilizer and the like, if necessary, can be added to the pharmaceutical preparation. The resulting pharmaceutical preparation can be sterilized by suitably carrying out treatment such as filtration through a bacterial filter, formulation of a disinfectant heating, irradiation or the like. An aseptic solid pharmaceutical preparation is produced and can be used by dissolving or suspending the pharmaceutical preparation in a suitable aseptic solution just before use.

[0055] Examples of the dosage form administered to eyes other than the eye drop include an ophthalmic ointment formed by using vaseline or the like; a liniment solution using a dilute iodine tincture, a zinc sulfate solution, a methylrosaniline chloride solution or the like; a dusting powder for directly administering a fine powder of an active ingredient; or an insert agent used by formulating or impregnating a suitable substrate or a material with an active ingredient and inserting the resultant substrate or material into palpebrae or the like.

[0056] A solution or a suspension of the active ingredient and a commonly used pharmaceutical vehicle is employed for inhalation and used as, for example an aerosol spray for inhalation. The active ingredient in the form of a dry powder can be administered even with an inhalator or other apparatuses so that the active ingredient can directly be brought into contact with the lungs.

[0057] The dose of the active ingredient of the cancer remedy of the present invention depends on the kinds of diseases, administration routes, conditions, ages, sexuality, body weight and the like of patients; however, the dose is usually about 1 to 1000 mg/day and is preferably formulated so as to satisfy the conditions.

[0058] As specifically described in Examples, the active ingredient of the cancer remedy of the present invention inhibits the growth of L929 cells having a strong growth property at a low concentration. Since the active ingredient is capable of similarly inhibiting even the growth of various human cultured cancer cells at a low concentration, the active ingredient is a very useful compound as a carcinostatic agent.

## 50 Examples

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**[0059]** The present invention will be explained specifically hereafter with Reference Examples and Examples. The groups of compounds used are described below; however, the present invention is not limited only to the Examples. In some cases, the <sup>1</sup>H-NMR peaks derived from carboxylic acids, hydroxy groups, amines and amides are not observed. There are some cases where an amine substance is a hydrochloride though not specifically mentioned.

[0060] When there is a description "the following compounds were synthesized according to the same method using the respective corresponding substrates", the reagents were synthesized using the substrates corresponding to the products. However, when it was difficult to understand, part of the substrates were also specified. In these reactions,

though there is a somewhat difference in the reaction temperature, reaction time and method for purification, it is needless to say that appropriate conditions can easily be found by trials if persons are those skilled in the art. The number (compound No.) after the generic name of the compound in each Example indicates the "compound No." listed in the above table.

[Example 1]

Preparation of methyl 2-(4-(2-naphthyloxy)benzamido)benzoate (compound No. 1)

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[0062] In 500 mL of dry methylene chloride, was suspended 29.1 g (0.11 mol) of 4-(2-naphthyloxy)benzoic acid under a nitrogen atmosphere. To the resulting suspension, was then added 15.4 g (0.121 mol) of oxalyl chloride. Ten drops of DMF were subsequently added to the suspension with a pipet. The mixture liquid was stirred at 35°C for 2 hours. The reaction liquid was then concentrated with an evaporator, and the residue was dissolved in 300 mL of dry methylene chloride. The resulting solution was dropped into a solution (250 mL) of 16.6 g (0.11 mol) of methyl anthranilate and 12.3 g (0.121 mol) of triethylamine in dry methylene chloride under cooling with ice under a nitrogen atmosphere. The mixture liquid was stirred under cooling with ice for 4 hours and then stirred at room temperature overnight. Water was added to the reaction liquid, and the resulting reaction liquid was extracted with methylene chloride twice. The organic layer was washed with a saturated brine and then dried over anhydrous sodium sulfate to distill off the solvent. The resulting residue was recrystallized from isopropyl alcohol (1.6 L) to provide 40.26 g (yield 92%) of methyl 2-(4-(2-naphthyloxy)benzamido)benzoate. Colorless needlelike crystals.

<sup>1</sup>H-NMR(CDCl<sub>3</sub>)  $\delta$  (ppm):

3.96 (S, 3H), 7.09-7.17 (m, 3H), 7.27-7.31 (m, 1H), 7.42-7.53 (m, 3H), 7.58-7.64 (m, 1H), 7.76 (d, J=8.5Hz, 1H), 7.84-7.90 (m, 2H), 8.03-8.10 (m, 3H), 8.93 (d, J=8.3Hz, 1H), 12.02 (br. s, 1H).

35 [Example 2]

Preparation of 2-(4-(2-naphthyloxy)benzamido)benzoic acid (compound No. 4)

[0063]

[0064] In a mixed solvent of methanol/THF (200 mL/400 mL), was dissolved 40.26 g (0.101 mol) of the methyl 2-(4-(2-naphthyloxy)benzamido)benzoate obtained in Example 1. To the resulting solution, was added 127 mL (0.51 mol) of a 4 M aqueous solution of lithium hydroxide. The mixture liquid was stirred at room temperature overnight. A 5 M hydrochloric acid was added to the reaction liquid to adjust the pH to about 1, and the reaction liquid was then stirred at room temperature for 0.5 hour. Water was added to the reaction liquid, and the obtained reaction liquid was extracted with ethyl acetate twice. The organic layer was washed with a saturated brine and then dried over anhydrous sodium sulfate, and the solvent was distilled off. The residue was recrystallized from isopropyl alcohol (1.3 L) to afford 31.23 g (yield 80%) of 2-(4-(2-naphthyloxy)benzamido)benzoic acid. Colorless needlelike crystals.

<sup>1</sup>H-NMR(CDCl<sub>3</sub>) δ (ppm):

7.12-7.18 (m, 3H), 7.27-7.30 (m, 1H), 7.43-7.53 (m, 3H), 7.65 (dt, J = 1.7 and 8.6Hz, 1H), 7.76 (d, J = 7.6Hz, 1H), 7.85-7.91 (m, 2H), 8.03 (dd, J = 2.0 and 6.9Hz, 2H), 8.14 (dd, J = 1.7 and 7.9Hz, 1H), 8.96 (d, J = 7.6Hz, 1H), 11.84 (br. s, 1H).

<sup>5</sup> [Examples 3 to 69]

[0065] In the following Examples, the compounds used in the present invention were prepared according to the method in Example 1 or 2 using the respective corresponding starting materials. The following tables show 1H-NMR spectral data and reaction yields of the prepared compounds. The compound No. in the tables corresponds to the compound No. listed in the tables mentioned above. The spectral data marked with "%" are measured data in DMSO-d<sub>6</sub>. All the others are data measured in CDCl<sub>3</sub>.

Example	Compound	<sup>1</sup> H-NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
3	10	3.85 (s, 3H), 7.11 (t, J = 8.3Hz, 1H), 7.20-7.30 (m, 3H), 7.40-7.65 (m, 5H), 7.70-7.90 (m, 4H), 8.05 (d, J = 7.3Hz, 1H), 8.88 (d, J = 9.2Hz, 1H),	66
4	11	$\frac{12.00 \text{ (s, 1H)}}{\text{(%) } 7.01 \text{ (t, J} = 7.4\text{Hz, 1H)}, 7.20-7.40 \text{ (m, }}$	65
		3H),7.40·7.55 (m, 3H), 7.58 (t, J = 7.9Hz, 1H), 7.69(s, 1H), 7.85 (d, J = 7.6Hz, 1H), 7.92 (d, J = 1.7Hz, 1H), 7.95 (s, 1H), 8.00 (d, J = 8.9Hz, 1H), 8.05 (d, J = 8.6Hz, 1H), 8.62 (d, J = 7.9Hz, 1H), 12.10 (br.s, 1H).	
5	12	3.95 (s, 3H), 7.05-7.20 (m, 2H), 7.25-7.35 (m, 2H), 7.45-7.60 (m, 3H), 7.70-7.90 (m, 2H), 7.93 (d, J = 1.3Hz, 1H), 8.11 (dt, J = 1.3 and 10.0Hz, 1H), 8.70 (d, J = 2.3Hz, 1H), 8.87 (d, J = 7.6Hz, 1H), 12.20 (s, 1H).	
6	14	(%) 7.24 (t, J = 7.2Hz, 1H), 7.34 (d, J = 8.5Hz,1H), 7.43 (dd, J = 2.6 and 4.5Hz, 1H), 7.50:7.60(m, 2H), 7.60-7.70 (m, 2H), 7.95 (dd, J = 8.1 and14.6Hz, 2H), 8.06 (dd, J = 7.6 and 8.0Hz, 2H), 8.20-8.25 (m, 2H), 8.60-8.70 (m, 2H), 12.30 (s, 1H).	34
7	18	(%) 6.95 (d, J = 8.6Hz, 1H), 7.16 (t, J = 7.6Hz, 1H), 7.43·7.62 (m, 4H), 7.73 (d, J = 2.3Hz, 1H), 7.88·8.07 (m, 4H), 8.55 (s, 1H), 8.70 (d, J = 7.9Hz, 1H).	55
8	56	3.77 (s, 2H), 3.89 (s, 3H), 7.05·7.11 (m, 3H), 7.27·7.57 (m, 7H), 7.69 (d, J = 7.6Hz, 1H), 7.82 (d, J = 8.6Hz, 2H), 8.01 (dd, J = 1.7 and 7.9Hz, 1H), 8.73 (d, J = 8.6Hz, 1H), 11.10 (br. s, 1H).	65

5	Example	Compound	¹ H·NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
J	9	59	3.80 (s, 2H), 7.04 (t, J = 7.6Hz, 1H), 7.09-7.14 (m, 2H), 7.21-7.29 (m, 2H), 7.34-7.45 (m, 4H),	81
			7.54-7.65 (m, 2H), 7.76 (d, J = 8.6Hz, 2H), 8.07	
10			(dd, $J = 1.7$ and $7.9$ Hz, 1H), $8.76$ (dd, $J = 1.0$	
			and 8.6Hz, 1H), 10.74 (br. s, 1H).	
	10	120	$3.96$ (s, 3H), $6.55$ (d, $J = 15.5$ Hz, 1H), $7.04 \cdot 7.13$	85
15		(trans)	(m, 3H), 7.25-7.31 (m, 1H), 7.41-7.52 (m, 3H),	
			7.56-7.62 (m, 3H), 7.72-7.78 (m, 2H), 7.83-7.89	
			(m, 2H), 8.06 (dd, J = 1.7 and 7.9Hz, 1H), 8.88	
20			(dd, J = 1.0 and 8.6Hz, 1H), 11.35 (br. s, 1H).	
	11	114	2.78  (t, J = 7.3 Hz, 2H), 3.09  (t, J = 7.3 Hz, 2H),	90
		,	3.92 (s, 3H), 6.99·7.03 (m, 2H), 7.05·7.12 (m,	
25			1H), 7.22·7.27 (m, 4H), 7.41 (dquint, J = 1.3	
			and 6.9Hz, 2H), 7.55 (dt, J = 1.7 and 6.9Hz,	
			1H), 7.67 (d, $J = 7.6Hz$ , 1H), 7.81 (d, $J = 8.6Hz$ ,	
			2H), 8.03 (dd, $J = 1.7$ and 7.9Hz, 1H), 8.73 (dd,	
30			J = 1.0 and 8.6Hz, 1H), 11.09 (br. s, 1H).	
	12	115	2.78  (t,  J = 7.9 Hz,  2 H), 3.09  (t,  J = 7.9 Hz,  2 H),	69
			6.97·7.02 (m, 2H), 7.12 (dt, J = 1.0 and 7.3Hz,	İ
35			1H), 7.20·7.27 (m, 4H), 7.41 (dquint, J = 1.3	
			and 6.9Hz, 2H), 7.57-7.68 (m, 2H), 7.80 (d, J =	
			8.9Hz, $2$ H), $8.10$ (dd, $J = 1.7$ and $7.9$ Hz, $1$ H),	
40			8.76 (dd, J = 1.0 and 8.6Hz, 1H), 10.87 (br. s,	
			1H).	
	13	121	6.54 (d, J = 15.5Hz, 1H), 7.05-7.08 (m, 2H),	84
45		(trans)	7.11·7.17 (m, 1H), 7.24·7.29 (m, 1H), 7.40·7.52	
			(m, 3H), 7.56·7.65 (m, 3H), 7.67·7.88 (m, 4H),	4
			8.15 (dd, $J = 1.7$ and 8.2Hz, 1H), 8.91 (dd, $J =$	1
[			1.0 and 8.6Hz, 1H), 11.16 (br. s, 1H).	
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	Example	Compound	<sup>1</sup> H·NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield
5				(%)
	14	126	$1.64 \text{ (d, } J = 7.3 \text{Hz, } 3\text{H), } 3.75 \cdot 3.83 \text{ (m, } 1\text{H), } 3.89$	65
			(s, 3H), 7.03·7.09 (m, 3H), 7.24·7.28 (m, 1H),	
10			$7.33 \cdot 7.56$ (m, 6H), $7.69$ (dd, $J = 1.7$ and $7.6$ Hz,	
			1H), $7.81$ (d, $J = 8.9$ Hz, 2H), $8.00$ (dd, $J = 1.7$	
			and 7.9Hz, 1H), $8.74$ (dd, $J = 1.0$ and $8.6$ Hz,	
			1H), 11.14 (br.s, 1H).	
15	15	127	( $\%$ ) 1.48 (d, J = 6.9Hz, 3H), 3.88 (q, J =	71
			6.9Hz, 1H), $7.04$ · $7.14$ (m, 3H), $7.28$ (dd, $J = 2.3$	
:			and 8.9Hz, 1H), $7.39-7.59$ (m, 6H), $7.80$ (d, $J = $	
20			7.6Hz, 1H), 7.90 (dd, $J = 1.3$ and 7.6Hz, 1H),	
			7.95  (d,  J = 8.2 Hz,  2 H),  8.52  (d,  J = 7.6 Hz,  1 H),	
			11.28 (br.s, 1H).	
25	16	128	1.73 (s, 6H), 3.84 (s, 3H), 7.07 (d, $J = 8.9 \text{ Hz}$ ,	88
			3H), 7.25-7.29 (m, 1H), 7.34-7.56 (m, 6H), 7.68	ļ
			(d, J = 7.9Hz, 1H), 7.81 (d, J = 8.9Hz, 2H), 7.99	
		:	(dd, $J = 1.7$ and $7.9$ Hz, 1H), $8.76$ (dd, $J = 1.0$	
30			and 7.6Hz, 1H), 10.95 (br.s, 1H).	
	17	129	( $\%$ ) 1.61 (s, 6H), 7.07 (d, J = 8.6Hz, 2H), 7.11	85
			(t, $J = 7.3$ Hz, 1H), $7.28$ (dd, $J = 1.6$ and $8.9$ Hz,	
35			1H), 7.40-7.52 (m, 5H), 7.58 (dt, J = 1.7 and	
			6.9Hz, 1H), 7.80 (d, J = 7.9Hz, 1H), 7.90-7.97	
			(m, 3H), $8.62$ (d, $J = 8.6$ Hz, 1H), $11.25$ (br.s,	
40			1H), 13.62 (br.s, 1H).	
	18	19	2.36 (s, 3H), 3.88 (s, 3H), 7.14 (d, J = 8.9Hz,	50
į			2H), 7.18-7.24 (m, 1H), 7.27-7.31 (m, 1H), 7.42-	
45			7.53 (m, 4H), 7.76 (d, J = 7.3Hz, 1H), 7.84-7.91	
45			(m, 3H), 8.01-8.04 (m, 2H), 10.18 (br.s, 1H).	
	19	20	2.36 (s, 3H), 7.13 (d, J = 8.6Hz, 2H), 7.21·7.30	71
			(m, 2H), $7.43-7.55$ (m, 4H), $7.76$ (d, $J = 7.3$ Hz,	
50			1H), $7.85-7.94$ (m, 3H), $8.00$ (d, $J = 8.9$ Hz, 2H),	Ì
			9.98 (br.s, 1H).	

	Example	Compound	<sup>1</sup> H-NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield
5				(%)
	20	74	3.80  (s, 2H), 3.95  (s, 3H), 7.10  (d, J = 6Hz, 2H),	53
i			$7.26 \cdot 7.46$ (m, 6H), $7.69$ (d, $J = 9Hz$ , 1H), $7.82$	
10			(d, $J = 9Hz$ , 2H), 7.87 (dd, $J = 2$ and 9Hz, 1H),	
70	,		8.17  (d,  J = 9Hz, 1H), 9.64  (d,  J = 2Hz, 1H),	
			11.12 (br.s, 1H).	
	21	75	(※) 3.83 (s, 2H), 7.08 (d, J = 8Hz, 2H), 7.29	69
15			(dd, J = 2 and 9Hz, 1H), 7.39-7.48 (m, 5H), 7.81	
			$(d, J = 8Hz, 1H), 7.89 \cdot 7.97 (m, 3H), 8.19 (d, J = )$	
			9Hz, 1H), 9.37 (d, J = 2Hz, 1H), 11.65 (br.s,	
20			1H).	
	22	76	3.77 (s, 2H), 3.88 (s, 3H), 6.77 (td, J = 2 and	62
			$7Hz$ , $1H$ ), $7.08$ (d, $J = 9Hz$ , $2H$ ), $7.31 \cdot 7.48$ (m,	
			6H), $7.69$ (d, $J = 8$ Hz, 1H), $7.82$ (d, $J = 9$ Hz,	
25			2H), 8.02 (dd, $J = 6$ and 9Hz, 1H), 8.57 (dd, $J = 1$	
		•	3 and 12Hz, 1H), 11.25 (br.s, 1H).	
ľ	23	77	3.78 (s, 2H), $6.68$ (m, 1H), $7.11$ (d, $J = 9$ Hz, 2H),	82
30			$7.20 \text{ (dd, J = 2 and 9Hz, 2H), } 7.32 \cdot 7.42 \text{ (m, 4H),}$	
			7.61 (d, J = 8Hz, 1H), 7.75 (d, J = 9Hz, 2H),	
ł			8.05 (t, $J = 6Hz$ , 1H), $8.56$ (dd, $J = 2$ and $12Hz$ ,	
35			1H), 10.88 (br.s, 1H).	
1	24	78	3.76 (s, 2H), 3.89 (s, 3H), 7.09 (d, J = 9Hz, 2H),	60
			7.26-7.45 (m, 7H), $7.68$ (dd, $J = 3$ and $9Hz$ , $2H$ ),	
			7.82  (d, J = 9Hz, 2H),  8.74  (dd, J = 5 and 9Hz,	
40			1H), 10.91 (br.s, 1H).	{
	25	79	(%) 3.75 (s, 2H), 7.07 (d, J = 9Hz, 2H), 7.29	85
			(dd, J = 3 and 9Hz, 1H), 7.38-7.48 (m, 6H), 7.54	
45			(dd, J = 3  and  9Hz, 1H), 7.82 (d, J = 8Hz, 1H),	
			7.88·7.91 (m, 1H), 7.95 (d, J = 9Hz, 1H), 8.52	
			(d, J = 9Hz, 1H), 11.99 (br.s, 1H).	
50				

	Example	Compound	¹H·NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
5	26	80	3.76 (s, 2H), 3.89 (s, 3H), 7.09 (d, J = 9Hz, 2H),	95
			$7.25 \cdot 7.51$ (m, 7H), $7.69$ (d, $J = 8Hz$ , 1H), $7.82$	
			(d, J = 9Hz, 2H), 7.98 (d, J = 3Hz, 1H), 8.73 (d, )	
10			J = 9Hz, 1H), 10.99 (br.s, 1H).	
	27	81	(%) 3.71 (s, 2H), 7.06 (d, J = 9Hz, 2H), 7.29	81
			(dd, J = 3 and 9Hz, 2H), 7.38-7.50 (m, 5H), 7.67	
15			(dd, $J = 3$ and $10Hz$ , $1H$ ), $7.82$ (d, $J = 8Hz$ , $1H$ ),	
			$7.89 \text{ (d, } J = 8Hz, 1H), 7.95 \text{ (d, } J = 9Hz, 1H),}$	
			8.50 (dd, J = 5 and 9Hz, 1H), 12.30 (br.s, 1H).	
20	28	82	2.41 (s, 3H), 3.73 (s, 2H), 3.81 (s, 3H), 6.96 (d, J	38
		i İ	= 8Hz, 1H), 7.10 (d, J = 9Hz, 2H), 7.27-7.46 (m,	
			7H), 7.70 (d, $J = 7Hz$ , 1H), 7.82 (dd, $J = 3$ and	ļ
			9Hz, $2$ H), $8.23$ (d, $J = 9$ Hz, $1$ H), $9.39$ (br.s, $1$ H).	
25	29	83	(%) 2.37 (s, 3H), 3.65 (s, 2H), 6.96 (d, J =	75
			7Hz, 1H), $7.04$ (d, $J = 8Hz$ , 2H), $7.17 \cdot 7.49$ (m,	
			7H), $7.73$ (d, $J = 8Hz$ , 1H), $7.81$ (d, $J = 8Hz$ ,	
30			1H), $7.90 \text{ (d, } J = 8Hz, 1H), } 7.94 \text{ (d, } J = 8Hz, $	
	J		1H), 10.57 (br.s, 1H).	
	30	238	3.93 (s, 3H), 3.95 (s, 3H), 7.09-7.24 (m, 6H),	45
35			7.42 (d, J = 2Hz, 1H), 7.61 (t, J = 7Hz, 1H),	
			7.67 (d, $J = 10Hz$ , 1H), 7.78 (d, $J = 9Hz$ , 1H),	}
			8.04 (d, J = 9Hz, 2H), 8.08 (dd, J = 2 and 8Hz,	
			1H), 8.93 (d, J = 9 Hz, 1H), 12.01 (br.s, 1H).	
40	31	239	3.98 (s, 3H), $7.08-7.26$ (m, 6H), $7.42$ (d, $J = 2Hz$ ,	67
			1H), 7.65-7.73 (m, 3H), 8.00-8.26 (m, 3H), 8.96	
			(d, J = 9Hz, 1H), 11.87 (br.s, 1H).	
45				

	Example	Compound	¹H-NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield
_				(%)
5	32	225	3.95 (s, 3H), 5.20 (s, 2H), 7.00·7.15 (m, 2H),	56
			7.20-7.30 (m, 4H), 7.35-7.45 (m, 4H), 7.49 (d, J	
			= 1.0Hz, 2H), 7.50·7.60 (m, 1H), 7.60·7.70 (m,	
10			1H), 7.76 (d, $J = 8.9$ Hz, 1H), 8.04 (dd, $J = 2.0$	
			and 9.9Hz, 2H), 8.10 (d, J = 1.7Hz, 1H), 8.90	
			(dd, J = 1.0 and 9.5Hz, 1H), 12.00 (br.s, 1H).	
15	33	226	(%) 5.23 (s, 2H), 7.17 (d, J = 8.7Hz, 2H),	82
			7.20·7.45 (m, 6H), 7.45·7.60 (m, 4H), 7.65 (t, J	
			= 7.5Hz, 1H), $7.82$ (d, $J = 8.9$ Hz, 1H), $7.90$ (d,	
20			J = 8.9Hz, 1H), 7.98 (d, $J = 8.9Hz$ , 2H), 8.05	
			(dd, J = 1.7  and  8.9 Hz, 1 H), 8.72 (d, J = 8.5 Hz,)	)
			1H), 12.20 (br.s, 1H), 13.70 (br.s, 1H).	
25	34	184	1.40 (s. 9H), 3.76 (s, 2H), 3.88 (s, 3H), 7.06 (d,	64
25			J = 8.6Hz, 3H), 7.09-7.23 (m, 2H), 7.32-7.38	
	,		(m, 4H), $7.53$ (t, $J = 7.3$ Hz, 1H), $7.60$ (d, $J =$	
			8.9Hz, $1$ H), $7.72$ (d, $J = 8.9$ Hz, $1$ H), $8.01$ (dd, $J$	
30			= 1.7 and $7.9$ Hz, $1$ H), $8.73$ (dd, $J = 1.0$ and	
			8.6Hz, 1H), 11.08 (br.s, 1H).	
	35	185	1.40 (s. 9H), 3.79 (s, 2H), 7.03·7.23 (m, 4H),	92
35			7.26·7.27 (m, 1H), 7.33·7.36 (m, 4H), 7.56 (t, J	
			= 8.9Hz, 2H), $7.69$ (d, $J = 8.9$ Hz, 1H), $8.08$ (d,	
			J = 8.3Hz, 1H), 8.76 (d, $J = 8.2Hz$ , 1H), 10.79	}
40			(br.s, 1H).	
	36	205	1.30-1.50 (m, 3H), 1.5-1.65 (m, 3H), 1.75-1.90	71
1			(m, 2H), 2.00-2.15 (m, 2H), 3.75 (s, 2H), 3.88	ļ
45			(s, 3H), 4.33·4.42 (m, 1H), 7.02·7.15 (m, 5H),	
}			7.20-7.24 (m, 1H), 7.31-7.37 (m, 3H), 7.53 (dt,	}
			J = 1.6 and $8.6$ Hz, 1H), $7.60$ (d, $J = 8.6$ Hz,	ļ
		į	1H), $7.68$ (d, $J = 8.9$ Hz, 1H), $8.00$ (dd, $J = 1.7$	
50			and 8.2Hz, 1H), 8.72 (dd, $J = 1.0$ and 8.6Hz,	İ
			1H), 11.07 (br.s, 1H).	

	Example	Compound	<sup>1</sup> H-NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield
5	37	206	1.25·1.65 (m, 6H), 1.75·1.90 (m, 2H), 2.00·2.15	(%) 86
	Ů.	_00	(m, 2H), 3.78 (s, 2H), 4.31-4.40 (m, 1H), 7.01-	
			7.13  (m, 6H), 7.18  (dd,  J = 1.6  and  8.9 Hz,  1 H),	
10			7.33 (d, $J = 8.6$ Hz, 2H), $7.55$ (d, $J = 9.9$ Hz, 2H),	
			7.63 (d, $J = 8.9$ Hz, 1H), $8.06$ (dd, $J = 1.7$ and	
			7.9Hz, 1H), $8.75$ (d, $J = 8.6$ Hz, 1H), $10.76$ (br.s,	
15	!		1H).	
	38	175	$0.89 \text{ (t, J = 6.9Hz, 3H), } 1.20 \cdot 1.45 \text{ (m, 8H), } 1.45 \cdot $	51
			1.65 (m, 2H), 1.84 (quint, $J = 6.6$ Hz, 2H), 3.75	
20			(s, 2H), $3.88$ (s, 3H), $4.06$ (t, $J = 6.6$ Hz, 2H),	
			7.03-7.15 (m, 5H), 7.21-7.25 (m, 1H), 7.32-7.37	
			(m, 3H), 7.53 (t, J = 7.3Hz, 1H), 7.60 (dd, J =	
25			2.3 and 7.6Hz, 1H), 7.69 (d, J = 8.9Hz, 1H),	
			8.01 (dd, $J = 1.7$ and 7.9Hz, 1H), 8.73 (dd, $J = 1.0$	
	00	150	1.0 and 8.3Hz, 1H), 11.08 (br.s, 1H).	
30	39	176	( $\%$ ) 0.85 (t, J = 6.6Hz, 3H), 1.25·1.55 (m,	86
			10H), 1.76 (quint, J = 6.6Hz, 2H), 3.75 (s, 2H), 4.05 (t, J = 6.6Hz, 2H), 7.01 (d, J = 8.6Hz, 2H),	
			7.10-7.15 (m, 2H), 7.23 (dd, $J = 2.3$ and 8.9Hz,	
35			1H), $7.32 \cdot 7.38$ (m, 4H), $7.57$ (t, $J = 7.3$ Hz, 1H),	
			7.72 (d, $J = 9.3Hz$ , 1H), $7.83$ (d, $J = 8.9Hz$ , 1H),	
į			7.95 (dd, $J = 1.7$ and 7.9 Hz, 1H), 8.50 (d, $J =$	
40		Ì	8.6Hz, 1H), 11.16 (br.s, 1H), 13.57 (br.s, 1H).	
	40	159	2.05 (s, 4H), 3.75 (s, 2H), 3.89 (s, 3H), 4.07 (t, J	70
1		ſ	= 5.6Hz, 2H), 4.15 (t, J = 5.6Hz, 2H), 6.87-6.97	
45			(m, 3H), 7.02-7.18 (m, 5H), 7.21-7.37 (m, 6H),	1
			7.50-7.57 (m, 1H), 7.60 (d, J = 9.6Hz, 1H), 7.69	j
			(d, $J = 8.9$ Hz, 1H), $8.01$ (dd, $J = 1.6$ and $7.9$ Hz,	
50			1H), 8.73 (dd, J = 1.0 and 8.6Hz, 1H), 11.08	
			(br.s, 1H).	

	Example	Compound	¹H·NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield
5				(%)
ŭ	41	160	(※) 1.80·2.00 (m, 4H), 3.72 (s, 2H), 4.04 (t, J	77
			= 5.6Hz, 2H), $4.14$ (t, $J = 5.6$ Hz, 2H), $6.88-6.94$	
			(m, 3H), 7.01 (d, J = 8.6Hz, 2H), 7.11-7.16 (m, 3H)	
10			2H), 7.21-7.30 (m, 3H), 7.35-7.38 (m, 4H), 7.53-	
			7.55  (m, 1H), 7.73  (d, J = 8.9 Hz, 1H), 7.82  (d, J)	
	] ]		= 8.9Hz, 1H), 7.94 (dd, J = 1.7 and 7.9Hz, 1H),	
15			8.49  (d,  J = 8.6 Hz,  1 H).	
	42	192	3.75 (s, 2H), $3.89$ (s, 3H), $4.65$ (d, $J = 6.3$ Hz,	47
			2H), 5.32 (d, J = 10.6Hz, 1H), 5.47 (d, J =	
20			17.5Hz, 1H), 6.05·6.20 (m, 1H), 7.03·7.23 (m,	
		į	6H), $7.32 \cdot 7.37$ (m, 3H), $7.53$ (t, $J = 7.3Hz$ , 1H),	
			7.61 (d, $J = 8.6$ Hz, 1H), 7.70 (d, $J = 8.9$ Hz, 1H),	
25			8.01 (dd, $J = 1.3$ and 7.9Hz, 1H), 8.73 (d, $J =$	
			8.6Hz, 1H), 11.08 (br.s, 1H).	
	43	193	( $\%$ ) 3.74 (s, 2H), 4.66 (d, J = 5.3Hz, 2H), 5.28	62
30		1	(dd, $J = 1.3$ and 10.6Hz, 1H), 5.44 (dd, $J = 1.7$	
30		j	and 17.5Hz, 1H), $6.03 \cdot 6.15$ (m, 1H), $7.02$ (d, $J = $	
			8.6Hz, 2H), 7.09-7.26 (m, 3H), 7.35-7.39 (m,	
		· 1	4H), $7.54-7.59$ (m, 1H), $7.74$ (d, $J = 8.9$ Hz, 1H),	
35	}		7.83 (d, $J = 8.9$ Hz, 1H), 7.95 (dd, $J = 1.3$ and	j
		1	8.2Hz, 1H), $8.50$ (d, $J = 8.6$ Hz, 1H), $11.15$ (br.s,	
}			1H), 13.56 (br.s, 1H).	
40	44		1.95 (quint, $J = 6.6$ Hz, 2H), 2.28 (q, $J = 6.9$ Hz,	54
			2H), 3.75 (s, 2H), 3.88 (s, 3H), 4.08 (t, J =	ļ
		1	6.6Hz, 2H), $5.02$ (dd, $J = 2.0$ and $10.3$ Hz, 1H),	
45	}	J	5.09  (dd,  J = 2.0  and  17.2Hz,  1H),  5.81-5.96  (m, )	
		1	1H), 7.03-7.16 (m, 5H), 7.21-7.25 (m, 1H), 7.32-	j
}		j	7.38 (m, 3H), 7.53 (dt, $J = 1.7$ and 7.3Hz, 1H),	
50			7.60  (d,  J = 9.6 Hz,  1 H), 7.69  (d,  J = 8.9 Hz,  1 H),	)
		l	8.00 (dd, $J = 1.7$ and 7.9Hz, 1H), 8.72 (dd, $J = 1.0$	
Ĺ			1.3 and 8.6Hz, 1H), 11.08 (br.s, 1H).	

5	Example	Compound	<sup>1</sup> H-NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
Ü	45	168	(※) 1.87 (quint, J = 6.3Hz, 2H), 2.23 (q, J =	89
			6.6Hz, 2H), $3.76$ (s, 2H), $4.08$ (t, $J = 6.6$ Hz, 2H),	
10			5.01 (d, J = 10.2Hz, 1H), 5.08 (dd, J = 2.0 and 17.2Hz, 1H), 5.82-5.97 (m, 1H), 7.03 (d, J =	
			8.6Hz, 2H), 7.11-7.17 (m, 2H), 7.25 (dd, $J = 2.6$	
			and 8.9Hz, 1H), 7.33·7.40 (m, 4H), 7.58 (t, J =	}
15			8.6Hz, 1H), 7.74 (d, J = 8.9Hz, 1H), 7.85 (d, J =	
			8.9Hz, $1$ H), $7.96$ (dd, $J = 1.7$ and $8.3$ Hz, $1$ H),	
			8.51 (dd, J = 8.3Hz, 1H), 11.14 (br.s, 1H), 13.54	
20			(br.s, 1H).	
	` 46	144	1.61 (s, 3H), 1.67 (s, 3H), 1.78 (s, 3H), 2.08-2.14	37
			(m, 4H), $3.75$ (s, 2H), $3.88$ (s, 3H), $4.64$ (d, $J =$	
25			6.6Hz, $2$ H), $5.11$ (br., $1$ H), $5.56$ (t, $J = 7.6$ Hz,	
		**	1H), 7.03-7.07 (m, 3H), 7.10-7.17 (m, 2H), 7.21-	
			7.25 (m, 1H), 7.32-7.37 (m, 3H), 7.53 (t, J =	
30			8.6Hz, 1H), 7.60 (d, $J = 9.6$ Hz, 1H), 7.70 (d, $J = 9.6$ Hz, 1H), 7.70 (d, $J = 9.6$ Hz, 1H)	Ì
ļ			8.9Hz, 1H), 8.01 (dd, $J = 1.7$ and 7.9Hz, 1H),	1
ĺ	47	145	8.72 (d, J = 8.3Hz, 1H), 11.07 (br.s, 1H). (※) 1.57 (s, 3H), 1.61 (s, 3H), 1.74 (s, 3H),	
35	41	i	2.02-2.13 (m, 4H), $3.76$ (s, 2H), $4.65$ (d, $J =$	66
			6.3Hz, 2H), 5.08 (br., 1H), 5.49 (t, J = 6.9Hz,	1
			1H), 7.03 (d, J = 8.6Hz, 2H), 7.11-7.16 (m, 2H),	
40		i	7.25 (dd, $J = 2.3$ and 8.9Hz, 1H), 7.35-7.40 (m,	
		)	4H), $7.58$ (t, $J = 8.6$ Hz, 1H), $7.73$ (d, $J = 8.9$ Hz,	
		1	1H), $7.83$ (d, $J = 8.9$ Hz, 1H), $7.96$ (d, $J = 1.7$ and	ì
45			7.9Hz, 1H), $8.51$ (d, $J = 7.9$ Hz, 1H), $11.14$ (br.s,	
10			1H), 13.50 (br.s, 1H).	

Example	Compound	¹ H-NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
48	233	3.75 (s, 2H), 3.88 (s, 3H), 5.17 (s, 2H), 7.02·7.11 (m, 3H), 7.20·7.26 (m, 3H), 7.32·7.56 (m, 9H), 7.62 (d, J = 9.6Hz, 1H), 7.70 (d, J = 8.9Hz, 1H), 8.01(dd, J = 1.7 and 8.2Hz, 1H), 8.73 (dd, J = 1.0 and 8.3Hz, 1H), 11.08 (br.s, 1H).	72
49	234	(※) 3.74 (s, 2H), 5.20 (s, 2H), 7.02 (d, J = 8.6Hz, 2H), 7.12 (t, J = 7.3Hz, 1H), 7.20-7.27 (m, 2H), 7.30-7.58 (m, 10H), 7.75 (d, J = 8.9Hz, 1H), 7.83 (d, J = 8.9Hz, 1H), 7.95 (dd, J = 1.3 and 7.9Hz, 1H), 8.50 (d, J = 7.9Hz, 1H).	78
50 -	223	2.17 (quint, J = 6.3Hz, 2H), 2.86 (t, J = 7.3Hz, 2H), 3.75 (s, 2H), 3.89 (s, 3H), 4.07 (t, J = 6.3Hz, 2H), 7.03-7.11 (m, 2H), 7.05 (d, J = 8.6Hz, 2H), 7.13-7.38 (m, 10H), 7.54 (dt, J = 1.7 and 7.3Hz, 1H), 7.61 (d, J = 8.9Hz, 1H), 7.67 (d, J = 8.9Hz, 1H), 8.00 (dd, J = 1.7 and 7.9Hz, 1H), 8.73 (dd, J = 1.0 and 8.6Hz, 1H), 11.08 (br.s, 1H).	59
51	224	(%) 2.03·2.14 (m, 2H), 2.79 (t, J = 7.3Hz, 2H), 3.76 (s, 2H), 4.07 (t, J = 6.3Hz, 2H), 7.03 (d, J = 8.3Hz, 2H), 7.11·7.40 (m, 12H), 7.52· 7.60 (m, 1H), 7.75 (d, J = 8.9Hz, 1H), 7.83 (d, J = 9.2Hz, 1H), 7.96 (dd, J = 1.7 and 7.9Hz, 1H), 8.51 (d, J = 8.6Hz, 1H), 11.18 (br.s, 1H).	86
52	136	3.75 (s, 2H), 3.88 (s, 3H), 5.34 (s, 2H), 7.03-7.10 (m, 3H), 7.22-7.28 (m, 3H), 7.33-7.37 (m, 3H), 7.46-7.71 (m, 6H), 7.84-7.90 (m, 3H), 7.94 (s, 1H), 8.00 (dd, J = 1.7 and 7.9Hz, 1H), 8.72 (d, J = 8.6Hz, 1H), 11.07 (br.s, 1H).	23

	Example	Compound	<sup>1</sup> H·NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
5	53	137	(※) 3.76 (s, 2H), 5.39 (s, 2H), 7.03 (d, J =	30
			$8.6$ Hz, 2H), $7.14$ (d, $J = 7.9$ Hz, 1H), $7.24 \cdot 7.30$ (m,	
			2H), $7.38$ (d, $J = 8.6$ Hz, 3H), $7.51 \cdot 7.66$ (m, 5H),	×
10			7.78  (d, J = 9.2Hz, 1H),  7.86  (d, J = 8.9Hz, 1H),	
!			7.93-7.98 (m, 4H), $8.05$ (s, 1H), $8.51$ (d, $J = 7.9$ Hz,	
			1H), 11.17 (br.s, 1H), 13.56 (br.s, 1H).	
15	54	21	3.94 (s, 3H), $7.11$ (t, $J = 7.3$ Hz, 1H), $7.35$ (d, $J = 1$	83
			8.3Hz, 2H), 7.49-7.63 (m, 4H), 7.79-7.89 (m, 3H),	
			7.94  (d,  J = 8.6 Hz,  2 H),  8.04  (d,  J = 1.3 Hz,  1 H),	
20			8.07 (dd, J = 1.7 and 8.3Hz, 1H), 8.91 (d, J =	
20			7.9Hz, 1H), 12.02 (br. s, 1H).	
	55	22	7.15 (t, $J = 7.6$ Hz, 1H), 7.34 (d, $J = 8.6$ Hz, 2H),	91
1			7.49·7.65 (m, 4H), 7.80·7.92 (m, 5H), 8.04 (s, 1H),	]
25			8.13 (dd, J = 2.0 and 8.3Hz, 1H), 8.93 (d, J =	
			8.3Hz, 1H), 11.84 (br. s, 1H).	
	56	27	3.92 (s, 3H), $4.20$ (s, 2H), $7.09$ (t, $J = 7.3$ Hz, 1H),	100
30			7.22-7.48 (m, 5H), 7.56-7.64 (m, 2H), 7.76-7.82	
.			(m, 3H), $7.99$ (d, $J = 7.9$ Hz, 2H), $8.06$ (dd, $J = 1.3$	
ļ			and 8.3Hz, 1H), 8.93 (d, J = 8.6Hz, 1H), 12.01	
35			(br. s, 1H).	
	57	29	4.22 (s, 2H), 7.15 (t, J = 8.3Hz, 1H), 7.24-7.50 (m,	79
			5H), 7.63-7.69 (m, 2H), 7.77-7.83 (m, 3H), 7.96 (d,	
		1	J = 8.6  Hz, 2H), 8.14  (dd,  J = 1.7  and  7.9 Hz, 1H),	
40			8.96 (d, J = 7.9Hz, 1H), 11.80 (br, s. 1 H).	
	58	1	3.95 (s, 3H), 5.26 (s, 2H), 7.09-7.15 (m, 1H), 7.20-	60
	}		7.27 (m, 2H), 7.34 (dt, $J = 1.3$ and 7.9Hz, 1H),	}
45			7.44 (dt, $J = 1.3$ and 7.9Hz, 1H), 7.57-7.65 (m,	
			3H), $7.72$ (d, $J = 8.3$ Hz, 1H), $7.77$ (d, $J = 8.6$ Hz,	
		į (	2H), 8.06-8.11 (m, 3H), 8.94 (d, J = 8.3Hz, 1H), 12.07 (br. s, 1H).	
50	<u>.</u>		12.01 (01. 8, 111).	

Example	Compound	¹H·NMR data (CDCl <sub>8</sub> ) δ (ppm)	Yield (%)
59	37	5.28 (s, 2H), 7.17-7.45 (m, 5H), 7.63-7.80 (m, 6H), 8.07 (d, $J = 8.6$ Hz, 2H), 8.15 (dd, $J = 1.7$ and 7.9Hz, 1H), 8.95-8.99 (m, 1 H), 11.90 (s, 1 H).	85
60	99	3.78 (s, 2H), 3.85 (s, 3H), 5.18 (s, 2H), 7.06 (t, J=7.9Hz, 1H), 7.19-7.23 (m, 2H), 7.30-7.36 (m, 1H), 7.40-7.55 (m, 6H), 7.70-7.78 (m, 3H), 7.99 (dd, J=1.7 and 8.2Hz, 1H), 8.70 (d, J=8.3Hz, 1H), 11.10 (br. s, 1H).	65
61	100	3.81 (s, 2H), 5.19 (s, 2H), 6.98 (t, J = 7.9Hz, 1H), 7.16-7.21 (m, 2H), 7.30-7.45 (m, 4H), 7.50-7.56 (m, 3H), 7.65-7.77 (m, 3H), 8.03 (dd, J = 1.7 and 7.9Hz, 1H), 8.74 (dd, J = 1.0 and 8.6 Hz, 1 H), 10.68 (br. s, 1 H).	76
62		3.96 (s, 3H), 4.26 (s, 2H), 7.12 (dt, J = 1.3 and 8.3Hz, 1H), 7.38-7.50 (m, 5H), 7.60 (dt, J = 1.7 and 8.6Hz, 1H), 7.69-7.80 (m, 4H), 7.95 (d, J = 2.0Hz, 1H), 7.98 (d, J = 1.7Hz, 1H), 8.08 (dd, J = 1.7 and 7,9Hz, 1H), 8.92 (d, J = 8.6Hz, 1H), 12.01 (br. s, 1H).	87
63		4.26 (s, 2H), 7.15 (t-like, 1H), 7.29-7.47 (m, 5H), 7.63-7.80 (m, 5H), 7.92-7.95 (m, 2H), 8.13 (dd, J = 1.7 and 7.9Hz, 1H), 8.93-8.96 (m, 1H), 11.84 (s, 1H).	22
64	40	3.99 (s, 3H), 7.17 (t, J = 8.6Hz, 1H), 7.55-7.68 (m, 3H), 7.92-8.02 (m, 6H), 8.12 (dd, J = 1.3 and 7.9Hz, 1H), 8.18-8.28 (m, 3H), 8.96 (d, J = 8.6Hz, 1H), 12.21 (br. s, 1H).	79
65	41	7.18-7.24 (m, 1H), 7.58-7.70 (m, 3H), 7.92-8.01 (m, 6H), 8.17 (d, J = 8.6Hz, 3H), 8.28 (s, 1H), 8.99 (d, J = 8.3Hz, 1H), 12.04 (br. s, 1H).	79

5	Example	Compound	<sup>1</sup> H·NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
	66	42	3.96 (s, 3H), 7.15 (t, J = 7.3Hz, 1H), 7.50-7.70 (m, 5H), 7.90-8.20 (m, 8H), 8.92 (d, J = 7.6Hz,	90
10			1H), 12.10 (s, 1H).	
10	67	43	( $\%$ ) 7.24 (t, J = 7.3Hz, 1H), 7.50-7.70 (m,	69
			5H), 7.95 (d, J = 8.3Hz, 2H), 8.00-8.10 (m, 5H),	
15			8.11 (d, J = 1.7Hz, 1H), 8.68 (d, J = 8.3Hz, 1H), 12.30 (s, 1H), 13.80 (br.s, 1H).	
	68	44	3.45 (s, 3H), 3.94 (s, 3H), 5.47 (s, 1H), 7.11 (t, J	100
			= 7.3Hz, 1H), 7.41·7.52 (m, 3H), 7.55·7.64 (m,	
20			3H), 7.73·7.85 (m, 4H), 8.00·8.09 (m, 3H), 8.92 (d, J = 8.3Hz, 1H), 12.01 (br. s,1H).	
	69	45	3.45 (s, 3H), 5.48 (s, 1H), 7.15 (t, J = 7.3Hz,	80
25			1H), 7.40-7.68 (m, 6H), 7.80-7.85 (m, 4H), 8.00	
			(d, J = 8.2Hz, 2H), 8.14 (d, J = 7.9Hz, 1H), 8.95 (d, J = 8.6Hz, 1H), 11.82 (br. s, 1H).	Ì

[Example 70]

Preparation of N-phenyl-(4-(2-naphthyloxy))benzamide (compound No. 5)

## [0066]

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 COOH +  $^{\text{H}_2\text{N}}$   $\bigcirc$   $\bigcirc$   $\bigcirc$   $\bigcirc$   $\bigcirc$   $\bigcirc$ 

[0067] In 5 mL of dry methylene chloride, was suspended 53 mg (0.20 mmol) of 4-(2-naphthyloxy)benzoic acid under a nitrogen atmosphere. To the resulting suspension, were then added 56 mg (0.44 mmol) of oxalyl chloride. One drop of DMF was subsequently added to the suspension with a pipet. The mixture liquid was stirred at 35°C for 1.5 hours. The reaction liquid was concentrated with an evaporator, and the residue was dissolved in 5 mL of dry methylene chloride. The resulting solution was then dropped into a dry methylene chloride solution (5 mL) of 19 mg (0.20 mmol) of aniline and 22 mg (0.22 mmol) of triethylamine under cooling with ice under a nitrogen atmosphere. The mixture liquid was stirred under cooling with ice for 4 hours and then at room temperature overnight. Water was added to the reaction liquid, and the resulting reaction liquid was extracted with methylene chloride twice. The organic layer was washed with a saturated brine and then dried over anhydrous sodium sulfate, and the solvent was distilled off. The residue was purified by silica gel column chromatography (hexane : ethyl acetate = 20 : 1) to provide 27 mg (yield 40%) of N-phenyl-(4-(2-naphthyloxy))benzamide. Colorless solid.

<sup>1</sup>H-NMR(CDCl<sub>3</sub>)  $\delta$  (ppm):

7.10-7.18 (m, 3H), 7.24-7.29 (m, 2H), 7.34-7.53 (m, 4H), 7.62-7.65 (m, 2H), 7.74-7.77 (m, 2H), 7.86-7.90 (m, 3H).

[Example 71]

Preparation of 2-(4-(2-naphthyloxy)benzamido)phenol (compound No. 6)

#### [0068]

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$$\bigcirc$$
 COOH +  $\bigcirc$  H<sub>2</sub>N  $\bigcirc$  H OH

[0069] In 5 mL of dry methylene chloride, was suspended 144 mg (0.54 mmol) of 4-(2-naphthyloxy)benzoic acid under a nitrogen atmosphere. To the resulting suspension, was then added 76 mg (0.60 mmol) of oxalyl chloride. One drop of DMF was subsequently added to the suspension with a pipet. The mixture liquid was stirred at 35°C for 1.5 hours. The reaction liquid was concentrated with an evaporator, and the residue was dissolved in 9 mL of dry methylene chloride. The obtained solution was dropped into a dry methylene chloride solution (6 mL) of 59 mg (0.54 mmol) of o-aminophenol and 3 mL of dry pyridine under cooling with ice under a nitrogen atmosphere. The resulting solution was stirred under cooling with ice for 1.5 hours and then at room temperature for 3 days. Water was added to the reaction liquid, and the resulting reaction liquid was extracted with methylene chloride twice. The organic layer was washed with a saturated brine and then dried over anhydrous sodium sulfate, and the solvent was distilled off. The residue was purified by silica gel column chromatography (hexane: ethyl acetate = 20:1 to 10:1) to afford 147 mg (yield 76%) of 2-(4-(2-naphthyloxy)benzamido)phenol. Colorless solid.

<sup>1</sup>H-NMR(CDCl<sub>3</sub>)  $\delta$  (ppm):

6.89-6.96 (m, 1H), 7.03-7.23 (m, 5H), 7.28-7.29 (m, 1H), 7.44-7.76 (m, 3H), 7.78-7.79 (d, J = 1.7Hz, 1H), 7.85-7.94 (m, 4H), 8.67 (s, 1H).

30 [Example 72]

Preparation of 2-(4-(2-naphthyloxy)benzamido)benzenesulfonamide (compound No. 7)

## [0070]

[0071] In 5 mL of dry methylene chloride, was suspended 132 mg (0.5 mmol) of 4-(2-naphthyloxy)benzoic acid under a nitrogen atmosphere. To the resulting suspension, was then added 70 mg (0.55 mmol) of oxalyl chloride. One drop of DMF was subsequently added to the suspension with a pipet. The obtained mixture liquid was stirred at 35°C for 2 hours. The reaction liquid was concentrated with an evaporator, and the residue was dissolved in 5 mL of dry methylene chloride. The resulting solution was dropped into a dry methylene chloride solution (4 mL) of 86 mg (0.5 mmol) of o-aminobenzenesulfonamide and 2 mL of dry pyridine under cooling with ice under a nitrogen atmosphere. The solution was stirred under cooling with ice for 4 hours and then at room temperature overnight. Water was added to the reaction liquid, and the resulting reaction liquid was extracted with methylene chloride twice. The organic layer was washed with a saturated brine and dried over anhydrous sodium sulfate, and the solvent was distilled off. The residue was recrystallized from a mixed solvent of benzene/ethyl acetate (8 mL/3 mL) to provide 112 mg (yield 54%) of 2-(4-(2-naphthyloxy)benzamido)benzenesulfonamide. Colorless granular crystals.

1H-NMR(DMSO-d<sub>6</sub>) δ (ppm):

7.23 (d, J = 8.9Hz, 2H), 7.26-7.38 (m, 2H), 7.46-7.68 (m, 4H), 7.90 (d, J = 7.9Hz, 2H), 7.97 (d, J = 8.6Hz, 3H), 8.04 (d, J = 9.2Hz, 1H), 8.46 (dd, J = 1.0 and 8.6Hz, 1H).

[Example 73]

Preparation of 2-(4-(2-naphthyloxy)benzamido)benzonitrile (compound No. 8)

[0072]

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[0073] In 5 mL of dry methylene chloride, was suspended 264 mg (1.0 mmol) of 4-(2-naphthyloxy)benzoic acid under a nitrogen atmosphere. To the resulting suspension, was then added 140 mg (1.1 mmol) of oxalyl chloride. One drop of DMF was subsequently added to the suspension with a pipet. The mixture liquid was stirred at 36°C for 2 hours. The reaction liquid was concentrated with an evaporator, and the residue was dissolved in 7 mL of dry methylene chloride. The resulting solution was dropped into a dry methylene chloride solution (5 mL) of 118 mg (1.0 mmol) of anthranilonitrile and 111 mg (1.1 mmol) of triethylamine under cooling with ice under a nitrogen atmosphere. The mixture liquid was stirred under cooling with ice for 4 hours and then at room temperature overnight. Water was added to the reaction liquid, and the resulting reaction liquid was extracted with methylene chloride twice. The organic layer was washed with a saturated brine and then dried over anhydrous sodium sulfate, and the solvent was distilled off. The residue was purified by silica gel column chromatography (hexane: ethyl acetate = 20 1 to 5:1) to afford 263 mg (yield 72%) of 2-(4-(2-naphthyloxy)benzamido)benzonitrile. Colorless solid.

<sup>1</sup>H-NMR(CDCl<sub>3</sub>) δ (ppm):

7.15 (d, J = 8.9Hz, 2H), 7.18-7.30 (m, 2H), 7.46-7.54 (m, 3H), 7.61-7.69 (m, 2H), 7.76-7.79 (m, 1H), 7.85-7.96 (m, 4H), 8.34 (br.s, 1H), 8.61 (d, J = 8.6Hz, 1H).

30 [Example 74]

Preparation of 2-(4-(2-naphthylthio)benzamido)benzonitrile (compound No. 23)

[0074]

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[0075] Procedures were carried out in the same manner as in Example 73 by using 280 mg (1.0 mmol) of 4-(2-naph-thylthio)benzoic acid to afford 104 mg (yield 27%) of the title compound.

1H-NMR(CDCl<sub>3</sub>) δ (ppm):

7.21 (t, J = 8.6Hz, 1H), 7.33 (d, J = 8.6Hz, 2H), 7.49-7.68 (m, 4H), 7.78-7.89 (m, 4H), 8.06 (d, J = 1.3Hz, 1H), 8.31 (br. s, 1H), 8.59 (d, J = 8.6Hz, 1H).

[Example 75]

Preparation of 1-(4-(2-naphthyloxy)benzamido)-2-(tetrazol-5-yl)benzene (compound No. 9)

5 [0076]

15 [0077] In 3 mL of dry DMF, were suspended 109 mg (0.30 mmol) of the 2-(4-(2-naphthyloxy)benzamido)benzonitrile obtained in Example 73, 48 mg (0.9 mmol) of ammonium chloride and 59 mg (0.9 mmol) of sodium azide. The resulting suspension was stirred at 80°C for 24 hours. To the reaction liquid, were added 5 mL of water and 5 mL of a 5 M hydrochloric acid, and the obtained reaction liquid was extracted with ethyl acetate twice. The organic layer was washed with a saturated brine and dried over anhydrous sodium sulfate, and the solvent was distilled off. The residue was recrystallized from 15 mL of acetonitrile to provide 92 mg (yield 75%) of 1-(4-(2-naphthyloxy)benzamido)-2-(tetrazol-5-yl)benzene. Colorless needlelike crystals.

<sup>1</sup>H-NMR(CD<sub>3</sub>OD)  $\delta$  (ppm):

7.15-7.21 (m, 2H), 7.27-7.35 (m, 2H), 7.43-7.53 (m, 3H), 7.57-7.63 (m, 1H), 7.78-8.01 (m, 4H), 8.14-8.19 (m, 2H), 8.76-8.81 (m, 1H).

[Example 76]

Preparation of 1-(4-(2-naphthylthio)benzamido)-2-(tetrazol-5-yl)benzene (compound No. 24)

30 [0078]

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[0079] Procedures were carried out in the same manner as in Example 75 by using 50 mg (0.13 mmol) of the 2-(4-(2-naphthylthio)benzamido)benzonitrile obtained in Example 74 as a raw material to afford 43 mg (yield 77%) of the title compound.

<sup>1</sup>H-NMR(DMSO-d<sub>6</sub>)  $\delta$  (ppm):

7.42 (t, J = 8.6Hz, 1H), 7.48 (d, J = 8.3Hz, 2H), 7.67-7.70 (m, 4H), 8.00-8.09 (m, 6H), 8.24 (d, J = 1.7Hz, 1H), 8.57 (d, J = 7.6Hz, 1H), 11.56 (br.s, 1H).

[Example 77]

Preparation of methyl 2-(3-amino-4-(2-naphthyloxy)benzamido)benzoate (compound No. 15)

5 [0080]

[0081] In 20 mL of ethyl acetate, was dissolved 350 mg (0.79 mmol) of the methyl 2-(4-(2-naphthyloxy)-3-nitroben-zamido)benzoate obtained in Example 5 (compound No. 12). To the resulting solution, was added 97 mg of a 10% Pd/C. The system was kept under a hydrogen atmosphere, and stirred at room temperature for 4 hours. The reaction liquid was filtered through Celite, and the resulting filtrate was concentrated. The residue was purified by silica gel column chromatography (hexane: ethyl acetate = 4:1 to 2:1) to provide 200 mg (yield 61%) of methyl 2-(3-amino-4-(2-naphthyloxy)benzamido)benzoate.

 $^{1}\text{H-NMR(CDCl}_{3}) \delta \text{ (ppm):}$ 

3.95 (s, 3H), 6.95 (d, J = 8.5Hz, 1H), 7.11 (t, J = 7.0Hz, 1H), 7.30 (dd, J = 2.3 and 8.9Hz, 2H), 7.36 (dd, J = 2.3 and 8.2Hz, 2H), 7.40-7.50 (m, 3H), 7.57 (d, J = 2.0Hz, 1H), 7.61 (dd, J = 1.4 and 8.6Hz, 1H), 7.71 (d, J = 7.9Hz, 1H), 7.84 (t, J = 8.0Hz, 2H), 8.07 (dd, J = 1.5 and 8.7Hz, 1H), 8.92 (d, J = 1.3 and 8.3Hz, 1H), 11.90 (s, 1H).

[Example 78]

Preparation of 2-(3-amino-4-(2-naphthyloxy)benzamido)benzoic acid (compound No. 16)

30 [0082]

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40 [0083] Procedures were carried out in the same manner as in Example 2 by using 200 mg (0.48 mmol) of the methyl 2-(3-amino-4-(2-naphthyloxy)benzamido)benzoate obtained in Example 77 as a raw material to afford 51 mg (yield 26%) of the title compound.

<sup>1</sup>H-NMR(DMSO-d<sub>6</sub>)  $\delta$  (ppm):

5.40 (br.s, 2H), 6.96 (d, J = 7.8Hz, 1H), 7.10-7.30 (m, 2H), 7.30-7.35 (m, 2H), 7.40-7.50 (m, 3H), 7.63 (dt, J = 1.5 and 7.6Hz, 1H), 7.82 (d, J = 7.8Hz, 1H), 7.90 (d, J = 7.8Hz, 1H), 7.96 (d, J = 9.8Hz, 1H), 8.05 (dd, J = 1.5 and 7.8Hz, 1H), 8.73 (d, J = 7.8Hz, 1H), 12.20 (s, 1H).

## [Example 79]

Preparation of 2-(4-(2-naphthyloxy)benzamido)benzoic acid (compound No. 4) sodium salt monoethanolate

### [0084]

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- [0085] In 250 mL of ethanol, was dissolved 10.35 mg (27.0 mmol) of the 2-(4-(2-naphthyloxy)benzamido)benzoic acid obtained in Example 2 with heating. To the resulting solution, was added 13.77 mL (27.54 mmol) of a 2 M aqueous solution of sodium hydroxide. The resulting solution was stirred at room temperature for 10 minutes and then allowed to stand overnight. The separated colorless solid was collected by filtration to provide 10.15 g (yield 83%) of the title compound.
  - <sup>1</sup>H-NMR(DMSO-d<sub>6</sub>) δ (ppm): 1.07 (t, J = 6.9Hz, 3H), 3.44-3.47 (m, 2H), 4.30-4.32 (m, 1H), 6.97 (t, J = 7.5Hz, 1H), 7.17 (d, J = 7.5Hz, 2H), 7.30 (t, J = 6.9Hz, 1H), 7.35 (d, J = 8.5Hz, 1H), 7.47-7.55 (m, 3H), 7.87 (d, J = 8.0Hz, 1H), 7.94 (d, J = 8.0Hz, 1H), 8.02 (t, J = 8.0Hz, 2H), 8.09 (d, J = 8.5Hz, 2H), 8.69 (d, J = 8.0Hz, 1H), 15.66 (br.s, 1H).

### 25 [Example 80]

Preparation of 2-(4-(2-naphthyloxy)benzamido)benzoic acid (compound No 4) lysine salt

### [0086]

[0087] In ethanol (6 mL), was dissolved 192 mg (0.5 mmol) of the 2-(4-(2-naphthyloxy)benzamido)benzoic acid obtained in Example 2. To the resulting solution, was added a methanol solution (3 mL) of 73 mg (0.5 mmol) of lysine (1-lysine, free base). The mixture liquid was stirred at room temperature for 5 minutes and then allowed to stand for 6 hours. The separated colorless solid was collected by filtration to afford 247 mg (yield 93%) of the title compound.  $^{1}$ H-NMR(CDCl<sub>3</sub>-CD<sub>3</sub>OD)  $\delta$  (ppm):

1.40-1.58 (m, 2H), 1.58-1.73 (m, 2H), 1.78-1.90 (m, 2H), 2.86-2.97 (m, 2H), 3.50-3.60 (m, 1H), 7.03-7.19 (m, 3H), 7.23-7.32 (m, 1H), 7.39-7.53 (m, 4H), 7.75-7,83 (m, 1H), 7.83-7.98 (m, 2H), 8.05-8.17 (m, 3H), 8.65-8.73 (m, 1H).

## [Example 81]

Preparation of 2-(4-(2-naphthyloxy)benzamido)benzoic acid (compound No 4) N-methyl-D-glucamine salt

#### [0088]

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[0089] In ethanol (12 mL), was dissolved 383 mg (1.0 mmol) of the 2-(4-(2-naphthyloxy)benzamido)benzoic acid obtained in Example 2. To the resulting solution, was added an aqueous solution (1 mL) of 195 mg (1.0 mmol) of N-methyl-D-glucamine. The mixture liquid was stirred at room temperature for 1 hour. The reaction liquid was filtered through a glass filter to remove fine insolubles. The filtrate was then concentrated. The residue thick malt syrupy substance was dissolved in a mixed solvent of 20 mL of water and 1 mL of methanol, and the obtained solution was freeze-dried to provide 542 mg (yield 94%) of the title colorless powdery compound.

1H-NMR(DMSO-d<sub>B</sub>) δ (ppm):

2.49-2.51 (m, 5H), 2.89-3.07 (m, 2H), 3.38-3.47 (m, 3H), 3.57-3.61 (m, 1H), 3.66-3.67 (m, 1H), 3.86 (br.s, 1H), 4.40-4.44 (br.s, 1H), 4.58 (br.s, 1H), 5.43 (br.s, 1H), 6.98 (t, J=8.6Hz, 1H), 7.20 (d, J=8.9Hz, 2H), 7.22-7.39 (m, 2H), 7.45-7.57 (m, 3H), 7.87-8.09 (m, 6H), 8.64 (d, J=8.3Hz, 1H).

#### [Example 82]

Preparation of 1-(4-(2-naphthyloxy)benzamido)-2-(tetrazol-5-yl)benzene (compound No. 9) sodium salt

## [0090]

[0091] In 80 mL of ethanol, was dissolved 732 mg (1.80 mmol) of the 1-(4-(2-naphthyloxy)benzamido)-2-(tetrazol-5-yl)benzene obtained in Example 75 with heating. To the resulting solution, was added 0.897 mL (1.80 mmol) of a 2 M aqueous solution of sodium hydroxide. The resulting mixture liquid was stirred at room temperature for 2.5 hours. The reaction liquid was concentrated, and the residue transparent film was dissolved in 30 mL of distilled water. The obtained solution was filtered through a filter (0.45  $\mu$  m), and the filtrate was freeze-dried to afford 767 mg (yield 99%) of the title colorless powdery compound.

### <sup>1</sup>H-NMR(DMSO-d<sub>6</sub>) $\delta$ (ppm):

7.15 (td, J = 1.5 and 7.8Hz, 1H), 7.25 (dt, J = 2.9 and 8.8Hz, 2H), 7.31 (td, J = 1.5 and 8.8Hz, 1H), 7.39 (dd, J = 2.5 and 8.8Hz, 1H), 7.47-7.54 (m, 2H), 7.60 (d, J = 2.4Hz, 1H), 7.90 (d, J = 7.8Hz, 1H), 7.90 (d, J = 7.8Hz, 1H), 8.03 (d, J = 9.3Hz, 1H), 8.25-8.30 (m, 3H), 8.79 (dd, J = 1.0 and 8.3Hz, 1H), 13.39 (br.s, 1H).

## [Example 83]

Preparation of 2-(4-(2-naphthyloxy)phenylacetamido)benzoic acid (compound No. 59) sodium salt

#### [0092]

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[0093] In 100 mL of ethanol, was dissolved 9.538 g (24.00 mmol) of the 2-(4-(2-naphthyloxy)phenylacetamido)) benzoic acid (compound No. 59) obtained in Example 9 with heating. To the resulting solution, was added 11.976 mL (24.00 mmol) of a 2 M aqueous solution of sodium hydroxide. The resulting mixture liquid was stirred at room temperature for 1.5 hours. The reaction liquid was concentrated, and the residue transparent film was dissolved in 200 mL of distilled water. The obtained solution was filtered through a filter (0.45 μm), and the filtrate was freeze-dried to provide 9.97 g (yield 99%) of the title colorless powdery compound.

 $^{1}\text{H-NMR}(\text{DMSO-d}_{6}) \delta \text{ (ppm)}$ :

3.65 (s, 2H), 6.95 (t, J = 8.2Hz, 1H), 7.10 (d, J = 8.6Hz, 2H), 7.25 (t, J = 7.3Hz, 1H), 7.33-7.36 (m, 1H), 7.37-7.53 (m, 5H), 7.93 (t, J = 7.3Hz, 2H), 7.99 (d, J = 8.9Hz, 2H), 8.46 (d, J = 8.3Hz, 1H), 14.80-14.91 (m, 1H).

### [Example 84]

Preparation of methyl 2-(4-(6-hydroxy-2-naphthyloxy)benzamido)benzoate (compound No. 213)

#### [0094]

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[0095] In 50 mL of THF, was dissolved 1.35 g (2.68 mmol) of the methyl 2-(4-(6-benzyloxy-2-naphthyloxy)benzamido))benzoate (compound No. 225) obtained in Example 32. To the resulting solution, was added 630 mg of a 10% Pd/C. The system was kept under a hydrogen atmosphere and stirred at room temperature for 32 hours. The reaction liquid was filtered through Celite, and the filtrate was concentrated to afford 1.04 g (yield 94%) of methyl 2-(4-(6-hydroxy-2-naphthyloxy)benzamido)benzoate.

<sup>1</sup>H-NMR(CDCl<sub>3</sub>)  $\delta$  (ppm):

3.88 (s, 3H), 5.26 (br.s, 1H), 6.90-7.20 (m, 6H), 7.35 (br.s, 1H), 7.50-7.70 (m, 3H), 7.90-8.05 (m, 3H), 8.84 (d, J = 7.6Hz, 1H), 11.95 (br.s, 1H).

#### [Example 85]

Preparation of 2-(4-(6-hydroxy-2-naphthyloxy)benzamido)benzoic acid (compound No. 214)

## 50 [0096]

[0097] Procedures were carried out in the same manner as in Example 2 by using 1.04 g (2.52 mmol) of the methyl 2-(4-(6-hydroxy-2-naphthyloxy)benzamido)benzoate (compound No. 213) obtained in Example 84 as a raw material to provide 0.78 g (yield 78%) of the title compound.

<sup>1</sup>H-NMR(DMSO-d<sub>6</sub>)  $\delta$  (ppm):

5 7.05-7.20 (m, 6H), 7.24 (s, 1H), 7.60 (dt, J = 2.0 and 9.0Hz, 1H), 7.74 (dd, J = 9.0 and 13.0Hz, 2H), 7.95 (d, J = 8.9Hz, 2H), 8.03 (dd, J = 1.7 and 8.0Hz, 1H), 8.28 (d, J = 9.0Hz, 1H), 9.70 (s, 1H), 12.20 (br.s, 1H), 13.70 (br.s, 1H).

[Example 86]

Preparation of methyl 2-(4-(6-hydroxy-2-naphthyloxy)phenylacetamido)benzoate (compound No. 217)

[0098]

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[0099] Procedures were carried out in the same manner as in Example 84 by using 50 mg (0.097 mmol) of the methyl 2-(4-(6-benzyloxy-2-naphthyloxy)phenylacetamido)benzoate (compound No. 233) obtained in Example 48 as a raw material to afford 22 mg (yield 53%) of the title compound.

<sup>1</sup>H-NMR(CDCl<sub>3</sub>)  $\delta$  (ppm):

3.76 (s, 2H), 3.89 (s, 3H), 5.26 (br.s, 1H), 7.02-7.15 (m, 5H), 7.22 (dd, J=2.3 and 8.9Hz, 1H), 7.31-7.37 (m, 3H), 7.53 (dt, J=1.7 and 8.9Hz, 1H), 7.60 (d, J=9.2Hz, 1H), 7.64 (d, J=8.9Hz, 1H), 8.01 (dd, J=1.7 and 8.3Hz, 1H), 8.72 (d, J=8.3Hz, 1H), 11.10 (br.s, 1H).

30 [Example 87]

Preparation of 2-(4-(6-hydroxy-2-naphthyloxy)phenylacetamido)benzoic acid (compound No. 218)

[0100]

[0101] Procedures were carried out in the same manner as in Example 2 by using 22 mg (0.05 mmol) of the methyl 2-(4-(6-hydroxy-2-naphthyloxy)phenylacetamido)benzoate (compound No. 217) obtained in Example 86 as a raw material to provide 9 mg (yield 42%) of the title compound.

<sup>1</sup>H-NMR (DMSO-d<sub>6</sub>)  $\delta$  (ppm):

3.83 (s, 2H), 7.07-7.28 (m, 6H), 7.41-7.46 (m, 3H), 7.65 (d, J = 7.6Hz, 1H), 7.75 (d, J = 8.9Hz, 1H), 7.81 (d, J = 8.9Hz, 1H), 8.04 (dd, J = 1.3 and 7.9Hz, 1H), 8.59 (d, J = 8.3Hz, 1H), 9.72 (s, 1H), 11.24 (br.s, 1H), 13.65 (br.s, 1H).

[Reference Example 1]

Synthesis of 4-(4-benzyloxyphenoxy)phenylacetic acid

### [0102]

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[0103] To hydroquinone monobenzyl ether (8.01 g, 40 mmol), were added benzene (100 mL) and methanol (25 mL). Into the resulting mixture, was slowly dropped 7.3 mL (38 mmol) of a 28% sodium methylate. The obtained mixture liquid was stirred at room temperature for 1 hour. The reaction liquid was concentrated, and pyridine (100 mL), 9.16 g (40 mmol) of methyl 4-bromophenylacetate and 1.25 g (12 mmol) of cupper(I) chloride were then added. The resulting mixture was stirred at 120°C for 30 hours with heating. The obtained reaction mixture was neutralized with hydrochloric acid, and the resultant product was extracted with ethyl acetate. The extract was dried and concentrated. The resulting concentrate was purified by silica gel chromatography to afford 4.76 g (13.7 mmol) of methyl ester of the objective compound.

[0104] In THF (10 mL), was dissolved 4.76 g (13.7 mmol) of the methyl ester compound. To the resulting solution, were added methanol (5 mL) and a 4 M aqueous solution (5 mL) of lithium hydroxide. The obtained reaction mixture liquid was stirred at room temperature for 4 hours. After completing the reaction, the resulting reaction liquid was neutralized with hydrochloric acid and concentrated until the quantity of the liquid reached a half. The produced crystals were collected by filtration and dried to provide 4.31 g (12.9 mmol) of the objective compound. Yield 94%.

[Example 88]

Preparation of methyl 2-(4-(4-benzyloxyphenoxy)phenylacetamido)benzoate (compound No. 315)

#### [0105]

[0106] To 4.30 g (12.9 mmol) of the 4-(4-benzyloxyphenoxy)phenylacetic acid obtained in Reference Example 1, were added methylene chloride (70 mL) and further 2.13 g (16.8 mmol) of oxalyl chloride under a nitrogen gas atmosphere. The resulting reaction mixture liquid was stirred at 50°C for 3 hours with heating. The obtained reaction liquid was concentrated, and the concentrate was dissolved in dry methylene chloride (60 mL). The resulting solution was cooled with ice, and 1.80 g (12.3 mmol) of methyl benzoate was then added to the cooled solution. To the obtained mixture, was further added 1.80 g (18.1 mmol) of triethylamine. The resulting mixture liquid was stirred at 50°C for 1 hour and further at room temperature overnight. The obtained reaction liquid was washed with water, and the reaction product was extracted with ethyl acetate. The extract was dried and concentrated. The obtained concentrate was purified by silica gel chromatography to afford 4.29 g (9.2 mmol) of the objective compound. Yield 75%.

1H-NMR(CDCI<sub>3</sub>) δ (ppm):

3.72(s, 2H), 3.87(s, 3H), 5.04(s, 2H), 6.91-7.02(m, 6H), 7.06(td, J = 8.6Hz, 1.6Hz, 1H), 7.24-7.46(m, 7H), 7.52(td, J = 8.0Hz, 1.6Hz, 1H), 7.99(dd, J = 8.2Hz, 1.6Hz, 1H), 8.71(dd, J = 8.6Hz, 1.3Hz, 1H), 11.03(brs, 1H)

[Examples 89 to 93]

[0107] Compounds listed in the following table (compound Nos. 313, 317, 319, 321 and 324) were synthesized

according to the same method as in Example 88. The table shows yields and results of NMR measurement of the compounds.

5	Example	Compound	<sup>1</sup> H-NMR data (CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
10	89	313	3.74 (2H, s), 3.85 (3H,s), 4.03 (2H, s), 6.91 (2H, d, J = 8.57Hz), 7.00 (2H, d, J = 8.58Hz), 7.06 (1H, ddd, J = 0.99, 7.25, 7.92Hz), 7.19-7.26 (7H, m), 7.34 (2H, d, J=8.25Hz), 7.52 (1H, ddd, J = 1.32, 7.26, 8.57Hz), 7.99 (1H, dd, J = 1.32, 7.91Hz), 8.72 (1H, d, J = 8.58Hz), 11.05 (1H, br).	63
	90	317	3.74 (2H, s), 3.84 (3H, s), 5.02 (2H, s), 6.62-6.74 (3H, m), 7.03 (2H, d, J = 8.6Hz), 7.21 (1H, t, J = 8.2Hz), 7.32 (2H, d, J = 8.6Hz), 7.37-7.40 (5H, m), 7.47 (1H, dd, J = 8.9 and 2.6Hz), 7.95 (1H, d, J = 2.6Hz), 8.72 (1H, d, J = 8.9Hz), 10.95 (1H, sbr).	46
15	91	319	3.75 (2H, s), 3.84 (3H, s), 5.02 (2H, s), 6.61-6.78 (4H, m), 7.03 (2H, d, J = 8.6Hz), 7.20 (1H, t, J = 8.2Hz), 7.33 (2H, d, J = 8.3Hz), 7.33-7.38 (5H, m), 7.97-8.03 (1H, m), 8.56 (1H, dd, J = 11.9 and 2.6Hz).	55
20	92	321	2.31 (3H, s), 3.73 (2H, s), 3.83 (3H, s), 5.01 (2H, s), 6.61-6.73 (3H, m), 7.02 (2H, d, J = 8.6Hz), 7.20 (1H, t, J = 8.2Hz), 7.31-7.41 (8H, m), 7.78 (1H, d, J = 2.0Hz), 8.60 (1H, d, J = 8.6Hz), 10.93 (1H, sbr).	47
25	93	324	3.72 (2H, s), 3.81 (3H, s), 5.07 (2H, s), 6.88-7.10 (7H, m), 7.14-7.28 (5H, m), 7.31 (2H, d, J = 8.25Hz), 7.50 (1H, ddd, J = 1.65, 7.26, 8.58Hz), 7.97 (1H, dd, J = 1.65, 7.92Hz), 8.72 (1H, d, J = 8.25Hz), 11.04 (1H, s).	89

#### [Example 94]

Preparation of 2-(4-(4-benzyloxyphenoxy)phenylacetamido)benzoic acid (compound No. 316)

## [0108]

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[0109] In THF (5 ml), was dissolved 278 mg (0.59 mmol) of the methyl 2-(4-(4-benzyloxyphenoxy)phenylacetamido) benzoate obtained in Example 88. To the resulting solution, were added methanol (5 mL) and a 4 M aqueous solution (2 mL) of lithium hydroxide. The resulting mixture liquid was stirred at room temperature for 2 hours. After completing the reaction, the obtained reaction liquid was neutralized with hydrochloric acid and concentrated until the quantity of the liquid reached a half. Crystals formed in the concentrate were collected by filtration and dried. The resulting crystals were further recrystallized from acetonitrile to provide 130 mg (0.29 mmol) of the objective compound. Yield 49%.

1H-NMR(CDCl<sub>3</sub>) δ (ppm):

3.74 (2H, s), 5.00 (2H, s), 6.87-6.99 (6H, m), 7.08 (1H, t, J = 7.5Hz), 7.24-7.43 (7H, m), 7.57 (1H, t, J = 7.5Hz), 8.07 (1H, d, J = 8.0 Hz), 8.75 (1H, d, J = 8.0 Hz), 10.77 (1H, brs).

[Example 95]

Preparation of methyl 2-(4-(4-hydroxyphenoxy)phenylacetamido)benzoate (compound No. 296)

[0110]

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[0111] In ethyl acetate (17 mL), was dissolved 4.20 g (9.0 mmol) of the methyl 2-(4-(4-benzyloxyphenoxy)phenylacetamido)benzoate obtained in Example 88 under a nitrogen gas atmosphere. A 10% palladium carbon (800 mg) was added to the resulting solution to prepare a reaction mixture, The nitrogen gas was replaced with hydrogen gas, and the reaction mixture was stirred at room temperature for 32 hours. The obtained reaction liquid was filtered on Celite and concentrated. The resulting concentrate was recrystallized from ethyl acetate to afford 2.26 g (6.0 mmol) of the objective compound. Yield 66%.

 $^{1}$ H-NMR(DMSO-d<sub>6</sub>)  $\delta$  (ppm):

3.70 (2H, s), 3.78 (3H, s), 6.76 (2H, d, J = 8.9Hz), 6.88 (4H, d-like, J = 8.6Hz), 7.18 (1H, t, J = 7.5Hz), 7.30 (2H, d, J = 8.6Hz), 7.59 (1H, J = 7.8Hz), 7.89 (1H, dd, J = 7.9Hz, 1.7Hz), 8.29 (1H, d, J = 7.6Hz), 9.31 (1H, s), 10.61 (1H, brs).

[Example 96]

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[0112] Procedures were carried out in the same manner as in Example 95 to synthesize compound No. 294. Yield 93%.

<sup>1</sup>H-NMR(CDCl<sub>3</sub>)  $\delta$  (ppm):

3.98 (s, 3H), 6.92 (d, 2H, J = 8.91Hz), 7.01-7.15 (m, 4H), 7.32 (t,1 H, J = 8.24Hz), 7.76 (t, 1H, J = 8.56Hz), 8.02 (d, 2H, J = 8.59Hz), 8.10 (dd, 1H, J = 1.32, 7.91Hz), 8.65 (d, 1H, J = 8.26Hz), 9.57 (s, 1H), 11.63 (s, 1H).

[Example 97]

Preparation of methyl 2-(4-(4-cyclohexyloxyphenoxy)phenylacetamido)benzoate (compound No. 287)

[0113]

[0114] To an N-methylmorpholine (4 mL) solution of 250 mg (0.66 mmol) of the methyl 2-(4-(4-hydroxyphenoxy) phenylacetamido)benzoate obtained in Example 95, were added 350 mg (1.3 mmol) of triphenylphosphine, 0.13 mL (1.3 mmol) of cyclohexanol and 230 mg (1.3 mmol) of diethyl azodicarboxylate under a nitrogen gas atmosphere. The resulting mixture liquid was stirred at room temperature for 2 hours. To the obtained mixture liquid, were further added 350 mg (1.3 mmol) of triphenylphosphine, 0.13 mL (1.3 mmol) of cyclohexanol and 230 mg (1.3 mmol) of diethyl azodicarboxylate. The resulting mixture liquid was stirred at room temperature for 2 hours. After completing the etherifying reaction, white precipitates formed in the reaction mixture were removed by filtration, and the filtrate was purified by silica gel column chromatography to provide 241 mg (0.53 mmol) of the objective compound. Yield 81%.  $^{1}$ H-NMR(CDCl<sub>2</sub>)  $\delta$  (ppm):

1.10-1.60 (6H, m), 1.79-1.84 (2H, brm), 1.96-2.04 (2H, brm), 3.72 (2 H, s), 3.87 (3H, s), 4.10-4.19 (1H, m), 6.86 (2H, d, J = 9.2Hz), 6.96 (4H, d, J = 8.3Hz), 7.06 (1H, t, J = 8.3Hz), 7.30 (2H, d, J = 8.6H z), 7.52 (1H, td, J = 8.6Hz, 1.7Hz), 7.99 (1H, dd, J = 8.3Hz, 1.7H z), 8.71 (1H, dd, J = 8.6Hz, 1.0Hz), 11.03 (1H, brs).

#### [Example 98]

Synthesis of 2-(4-(4-cyclohexyloxyphenoxy)phenylacetamido)benzoic acid (compound No. 288)

### 5 [0115]

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[0116] In THF (8 mL), was dissolved 240 mg (0.53 mmol) of the methyl 2-(4-(4-cyclohexyloxyphenoxy)phenylacetamido)benzoate obtained in Example 97. To the resulting solution, were added methanol (5 mL) and a 4 M aqueous solution (2 mL) of lithium hydroxide. The obtained reaction mixture liquid was stirred at room temperature for 3 hours. After completing the hydrolysis reaction, the resulting reaction liquid was neutralized with hydrochloric acid and concentrated until the quantity of the liquid reached a half. The reaction product was extracted from the concentrate with ethyl acetate, and the extract was dried and concentrated. The obtained oily concentrate was recrystallized from acetonitrile to afford the objective compound (121 mg). Yield 51%.

 $^{1}H-NMR(DMSO-d_{6}) \delta (ppm)$ :

1.24-1.55 (6H, m), 1.65-1.75(2H, brm), 1.85-1.95 (2H, brm), 3.72 (2H, s), 4.20-4.28 (1H, m), 6.89-6.96 (6H, m), 7.13 (1H, t, J = 8.5Hz), 7.32 (2H, d, J = 8.6Hz), 7.56 (1H, t, J = 8.0Hz), 7.95 (1H, dd, J = 7.9Hz,1.7Hz), 8.61 (1H, d, J = 8.3Hz), 11.16 (1H, brs).

### [Examples 99 to 111]

[0117] The compounds listed in the following tables were synthesized according to the same method as in Example 97. The yields of the respective compounds were calculated on the basis of the molar amounts of the raw material hydroxy substances corresponding to the compounds.

05	Example	Compound	¹H-NMR(CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
<i>35</i> <i>40</i>	99	291	1.56-1.68(m, 2H), 1.76-1.88 (m,6H), 3.72 (s, 2H), 3.87 (s, 3H), 4.70 (brm, 1H), 6.83 (d, 2H, J = 8.9Hz), 6.96 (d, 2H, J = 8.6Hz), 6.97 (d, 2H, J = 8.9Hz), 7.05 (ddd, 1H, J = 7.9Hz, 6.9Hz, 1.0Hz), 7.30 (d, 2H, J = 8.6Hz), 7.51 (ddd, 1H, J = 8.6Hz, 6.9Hz, 1.3Hz), 7.98 (dd, 1H, J = 7.9Hz, 1.3Hz), 8.72 (dd, 1H, J = 8.6Hz, 1.0Hz), 11.03 (brs, 1H).	94
45	100	285	1.28-1.48 (m, 16H), 1.58-1.81 (m, 4H), 2.02-2.14 (m, 2H), 3.72 (s,2H), 3.86 (s, 3H), 4.34 (m, 1H), 6.84 (d, 2H, J = 9.2Hz), 6.96 (d, 2H, J = 9.2Hz), 6.97 (d, 2H, J = 8.6Hz), 7.05 (ddd, 1H, J = 7.9Hz, 6.9Hz, 1.0Hz), 7.30 (d, 2H, J = 8.6Hz), 7.51 (ddd, 1H, J = 8.6Hz, 6.9Hz, 1.7Hz), 7.98 (dd, 1H, J = 7.9Hz, 1.7Hz), 8.72 (d, 1H, J=8.6Hz), 11.03(s, 1H).	47
50	101	272	1.32 (d, 6H, J = 6.3Hz), 3.72 (s, 2H), 3.87 (s, 3H), 4.47 (sep, 1H, J = 6.3Hz), 6.84 (d, 2H, J = 8.9Hz), 6.96 (d, 2H, J = 8.6Hz), 6.97 (d, 2H, J = 8.9Hz), 7.06 (ddd, 1H, J = 8.2Hz, 6.9Hz, 1.0Hz), 7.30 (d, 2H, J = 8.6Hz), 7.51 (ddd, 1H, J = 8.6Hz, 6.9Hz, 1.7Hz), 7.98 (dd, 1H, J = 8.2Hz, 1.7Hz), 8.72 (dd, 1H, J = 8.6Hz, 1.0Hz), 11.03 (s, 1H).	72
55	102	265	0.96 (t, 6H, J = 7.6Hz), 1.67 (dq, 4H, J = 7.6Hz, 5.9Hz), 3.72 (s,2H), 3.86 (s, 3H), 4.03 (quint, 1H, J = 5.9Hz), 6.85 (d, 2H, J = 9.2Hz), 6.94-6.99 (m, 4H), 7.05 (ddd, 1H, J = 7.9Hz, 6.9Hz, 1.0Hz), 7.30 (d, 2H, J = 8.6Hz), 7.51 (ddd, 1H, J = 8.6Hz, 6.9Hz, 1.7Hz), 7.98 (dd, 1H, J = 7.9Hz, 1.7Hz), 8.72 (dd, 1H, J = 8.6Hz, 1.0Hz), 11.03 (s, 1H).	50

## (continued)

	Example	Compound	<sup>1</sup> H-NMR(CDCl <sub>3</sub> ) δ (ppm)	Yield (%)
5	103	274	1.5-1.7 (8H, m), 1.77-1.95 (6H, brm), 3.95 (3H, s), 4.38 (1H, m), 6.88 (2H, d, J = 8.9Hz), 6.98-7.04 (4H, m), 7.11 (1H, t, J= 7.7Hz), 7.60 (1H, td, J = 7.6Hz, 1.6Hz), 8.00 (2H, d, J = 8.9Hz), 8.07 (1H, dd, J = 7.9Hz, 1.7Hz), 8.92 (1H, dd, J = 8.6Hz, 0.7Hz), 11.98 (1H, br).	82
10	104	276	1.51-1.74 (8H, brm), 1.76-2.00 (6H, brm), 3.96 (3H, s), 4.40 (1H, quint, J = 4.0Hz), 6.91 (2H, d, J = 8.9Hz), 7.00-7.06 (3H, m), 7.15 (1H, td, J = 7.6 and 1.3Hz), 7.62 (1H, td, J = 7.6 and 1.3Hz), 8.07-8.12 (2H, m), 8.63 (1H, d, J = 2.3Hz), 8.86 (1H, d, J = 8.0Hz), 12.15 (1H, brs).	85
15	105	281	1.40-1.60 (8H, m), 1.72-1.95 (6H, m), 2.31 (3H, s), 3.71 (2H, s), 3.86 (3H, s), 4.34 (1H, quint, $J = 4.0Hz$ ), 6.81 (2H, d, $J = 8.9Hz$ ), 6.94-6.98 (4H, m), 7.30 (2H, d, $J = 8.9Hz$ ), 7.34 (1H, m), 7.79 (1H, d, $J = 1.7Hz$ ), 8.59 (1H, d, $J = 8.6Hz$ ), 10.90 (1H, brs).	44
20	106	259	1.02 (6H, d, $J = 6.6Hz$ ), 2.07 (1H, quint, $J = 6.6Hz$ ), 3.69 (2H, d, $J = 6.6Hz$ ), 3.72 (2H, s), 3.88 (3H, s), 6.86 (2H, d, $J = 9.2Hz$ ), 6.95 (2H, d, $J = 7.9Hz$ ), 6.98 (2H, d, $J = 8.9Hz$ ), 7.06 (1H, td, $J = 7.7$ and 1.0Hz), 7.29 (2H, t, $J = 8.0Hz$ ), 7.52 (1H, td, $J = 7.9$ and 1.6Hz), 7.99 (1H, dd, $J = 7.9$ and 1.6Hz), 8.71 (1H, d, $J = 7.9Hz$ ).	46
25	107	284	1.23-1.92 (14H, m), 3.71 (2H, s), 3.87 (3H, s), 4.35 (1H, tt, $J = 3.96$ , 7.92Hz), 6.86-6.97 (4H, m),7.02-7.11 (3H, m), 7.28 (2H, m), 7.51 (1H. ddd, $J = 1.65$ , 7.26, 8.57Hz), 7.98 (1H, dd, $J = 1.65$ , 7.91Hz), 8.70 (1H, dd, $J = 0.99$ , 8.58Hz), 11.30 (1H, s).	69
30	108	257	3.72 (2H, s), $3.87$ (3H, s), $4.31$ (4H, s), $6.90$ - $7.02$ (10H, m), $7.06$ (1H, ddd, J = $8.2$ Hz, $7.3$ Hz, and $1.0$ Hz), $7.25$ - $7.33$ (4H, m), $7.52$ (1H, ddd, J = $8.6$ Hz, $7.3$ Hz and $1.7$ Hz), $7.99$ (1H, dd, J = $8.2$ Hz and $1.7$ Hz), $8.71$ (1H, dd, J = $8.6$ Hz and $1.0$ Hz), $11.04$ (1H, brs).	68
35	109	261	2.52(1H, ddd, $J = 6.9$ Hz, $6.6$ Hz, and $1.3$ Hz), $2.55$ (1H, ddd, $J = 6.9$ Hz, $6.9$ Hz, and $1.3$ Hz), $3.72$ (2H, s), $3.87$ (3H, s), $3.99$ (2H, t, $J = 6.9$ Hz), $5.11$ (1H, dd, $J = 10.2$ Hz, and $1.7$ Hz), $5.17$ (1H, dd, $J = 17.1$ Hz, and $1.7$ Hz), $5.90$ (1H, dddd, $J = 17.1$ Hz, $10.2$ Hz, $6.9$ Hz, and $6.6$ Hz), $6.86$ (2H, d, $J = 8.9$ Hz), $6.95$ (2H, d, $J = 8.6$ Hz), $6.98$ (2H, d, $J = 8.9$ Hz), $7.06$ (1H, ddd, $J = 8.2$ Hz, $7.3$ Hz, and $1.0$ Hz), $7.30$ (2H, d, $J = 8.6$ Hz), $7.51$ (1H, ddd, $J = 8.6$ Hz, $3.2$ Hz, and $3.2$ Hz, $3.2$ Hz, and $3.2$ Hz, $3.2$ Hz, and $3.2$ Hz, $3.2$ H	52
40	110	255	0.93 (6H, t, J = 7.3Hz), 1.34-1.55 (4H, m), 1.62-1.71 (1H, m), 3.72 (2H, s), 3.82 (2H, d, J = 5.6Hz), 3.87 (3H, s), 6.86 (2H, d, J = 8.9Hz), 6.95 (2H, d, J = 8.6Hz), 6.98 (2H, d, J = 8.9Hz), 7.06 (1H, ddd, J = 8.2Hz, 6.9Hz, and 1.0Hz), 7.30 (2H, d, J = 8.6Hz), 7.52 (1H, ddd, J = 8.6Hz, 6.9Hz, and 1.3Hz), 7.99 (1H, dd, J = 8.2Hz, and 1.3Hz), 8.71 (1H, dd, J = 8.6Hz, and 1.0Hz), 11.03(1H, brs).	41
45 50	111	269	0.98 (3H, t, J = 7.3Hz), 1.45-1.64 (2H, m), 1.71-1.82 (2H, m), 3.72 (2H, s), 3.87 (3H, s), 3.93 (2H,t, J = 6.6Hz), 6.85 (2H, d, J = 8.9Hz), 6.96 (2H, d, J = 8.6Hz), 6.98 (2H, d, J = 8.9Hz), 7.06 (1H,ddd, J = 8.2Hz, 6.9Hz, and 1.0Hz), 7.30 (2H, d, J = 8.6Hz), 7.52 (1H, ddd, J = 8.6Hz, 6.9Hz, and 1.7Hz), 7.99 (1H, dd, J = 8.2Hz, and 1.3Hz), 8.71 (1H, dd, J = 8.6Hz, and 1.0Hz), 11.03 (1H, brs).	95

[Examples 112 to 133]

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[0118] The following compounds listed in the following tables were synthesized according to the same method as in Example 98. The yields of the respective compounds were calculated on the basis of the molar amounts of the raw material methyl esters corresponding to the compounds.

	Example	Compound	<sup>1</sup> H-NMR (DMSO-d <sub>6</sub> ) δ (ppm):	Yield (%)
5	112	314	3.74 (2H, s), 4.16 (2H, s), 6.93 (2H, d, J = 8.91Hz), 6.97 (2H, d, J = 8.58Hz), 7.12 (1H, ddd, J = 1.32, 7.26, 7.92Hz), 7.19-7.38 (9H, m), 7.56 (1H, ddd, J = 1.65, 7.26, 8.58Hz), 7.94 (1H, dd, J = 1.65, 7.92Hz), 8.50 (1H, dd, J = 0.66, 8.25Hz), 11.14 (1H, s), 13.55 (1H, br).	78
10	113	292	1.53-1.85 (m, 6H), 1,86-1.92 (m,2H), 3.72 (s, 2H), 4.76 (br, 1H), 6.88-6.98 (m, 6H), 7.13 (dd, 1H, J = 7.9Hz, 7.6Hz), 7.32 (d, 2H, J = 8.6Hz), 7.57 (ddd, 1H, J = 8.9Hz, 7.6Hz, 1.3Hz), 7.95 (dd, 1H, J = 7.9Hz, 1.3Hz), 8.50 (d, 1H, J = 8.9Hz), 11.16 (s, 1H).	66
15	114	279	1.50-1.70 (8H, br), 1.70-1.90 (6H, brm), 3.72 (2H, s), 4.41 (1H, m), 6.87-6.98 (6H, m), 7.13 (1H, t, J = 6.9Hz), 7.32 (2H, d, J = 8.6Hz), 7.58 (1H, td, J = 7.9Hz, 1.7Hz), 7.95 (1H, dd, J = 7.9Hz, 1.7Hz), 8.50 (1H, d, J = 8.6Hz), 11.10 (1H, brs).	80
20	115	286	1.34-1.72 (m, 22H), 3.72 (s, 2H), 4.37 (br, 1H), 6.91 (d, 2H, J = 9.2Hz), 6.92 (d, 2H, J = 8.6Hz), 6.96 (d, 2H, J = 9.2Hz), 7.13 (dd, 1H, J = 7.9Hz, 7.3Hz), 7.33 (d, 2H, J = 8.6Hz), 7.57 (ddd, 1H, J = 8.6Hz, 7.3Hz, 1.7Hz), 7.95 (dd, 1H, J = 7.9Hz, 1.7Hz), 8.51 (d, 1H, J = 8.6Hz), 11.17 (s, 1H).	68
25	116	273	1.32 (d, 6H, $J = 6.3Hz$ ), 3.72 (s, 2H), 3.87 (s, 3H), 4.47 (sep, 1H, $J = 6.3Hz$ ), 6.84 (d, 2H, $J = 8.9Hz$ ), 6.96 (d, 2H, $J = 8.6Hz$ ), 6.97 (d, 2H, $J = 8.9Hz$ ), 7.06 (ddd, 1H, $J = 8.2Hz$ , 6.9Hz, 1.0Hz), 7.51 (ddd, 1H, $J = 8.6Hz$ , 6.9Hz, 1.7Hz), 7.98 (dd, 1H, $J = 8.2Hz$ , 1.7Hz), 8.72 (dd, 1H, $J = 8.6Hz$ , 1.0Hz), 11.22 (s, 1H), 13.56 (brs, 1H).	45
30	117	266	0.90 (d, J = 7.3Hz), 1.60 (dq, 4H, J = 7.3Hz, 5.9Hz), 3.72 (s, 2H), 4.14 (quint, 1H, J = 5.9Hz), 6.91 (d, 2H, J = 8.6Hz), 6.94 (s, 4H), 7.13 (dd, 1H, J = 7.9Hz, 7.6Hz), 7.32 (d, 2H, J = 8.6Hz), 7.57 (dd, 1H, J = 8.6Hz, 7.9Hz), 7.95 (d, 1H, J = 7.6Hz), 8.50 (d, 1H, J = 8.6Hz), 11.14 (s, 1H).	40
35	118	275	1.50-1.70(8H, br), 1.71-2.00 (6H, brm), 4.46 (1H, m), 6.95 (2H, d, $J = 9.2Hz$ ), 7.05-7.09 (4H, m), 7.20 (1H, t, $J = 7.7Hz$ ), 7.66 (1H, t, $J = 8.0Hz$ ), 7.94 (2H, d, $J = 8.9Hz$ ), 8.05 (1H, dd, $J = 7.9Hz$ , 0.7Hz), 8.70 (1H, d, $J = 8.6Hz$ ), 12.13 (1H, brs).	78
33	119	297	3.92 (s, 2H), 6.98 (d, 2H, J = 9.23Hz), 7.11-7.00 (m, 4H), 7.34 (t,1H, J = 7.59Hz), 7.51 (d, 2H, J= 8.24Hz), 7.77 (t, 1H, J= 8.24Hz), 8.16 (dd, 1H, J = 1.32, 7.91Hz), 8.71 (d, 1H, J = 8.24Hz), 9.53 (s, 1H), 11.34(s, 1H), 13.77 (br, 1H).	76
40	120	295	6.92 (d, 2H, J = 8.89Hz), 7.14-7.06 (m, 4H), 7.28 (t, 1H, J = 7.59Hz), 7.74 (t, 1H, J = 8.26Hz), 8.02 (d, 2H, J = 8.59Hz), 8.14 (dd, 1H, J = 1.32, 7.91Hz), 8.78 (d, 1H, J = 8.26Hz), 9.56 (s, 1H), 12.20 (s, 1H), 13.86 (br, 1H).	78
45	121	277	$ (*) 1.4-1.70 (8H, brm), 1.80-2.00 (6H, brm), 4.48 (1H, m), 6.99 (2H, d, J=9.2Hz), \\ 7.11-7.17 (3H, m), 7.23 (1H, t, J=7.9Hz), 7.67 (1H, t, J=8.3Hz), 8.05 (1H, d, J=7.5Hz), 8.16 (1H, dd, J=8.9 and 2.3Hz), 8.58-8,63 (2H, m), 12.23 (1H, brs). $	80
50	122	318	3.76 (2H, s), $5.07$ (2H, s), $6.55$ (1H, dd, J= $7.7$ and $1.8$ Hz), $6.64$ (1H, t, J= $2.3$ Hz), $6.79$ (1H, dd, J= $8.3$ and $1.8$ Hz), $6.98$ (2H, d, J= $8.6$ Hz), $7.27$ (1H, t, J= $8.3$ Hz), $7.31-7.42$ (7H, m), $7.63$ (1H, dd, J= $8.9$ and $2.6$ Hz), $7.88$ (1H, d, J= $2.6$ Hz), $8.53$ (1H, d, J= $2.6$ Hz), $11.05$ (1H, brs), $13.91$ (1H, brs).	66
50	123	280	1.40-1.60 (8H, brm), 1.72-1.96 (6H, m), 3.75 (2H, s), 4.31 (1H, quint, J= $4.0$ Hz), 6.78 (2H, d, J = $9.0$ Hz), 6.92 (2H, d, J = $9.0$ Hz), 6.98 (2H, d, J = $8.6$ Hz), 7.27 (2H, d, J = $8.6$ Hz), 7.52 (1H, dd, J = $9.2$ and $2.6$ Hz), $8.0$ 7 (1H, d, J = $2.6$ Hz), 8.75 (1H, d, J = $9.2$ Hz), 10.70 (1H, brs).	80

## (continued)

	Example	Compound	<sup>1</sup> H-NMR (DMSO-d <sub>6</sub> ) δ (ppm):	Yield (%)
5	124	320	3.81 (2H, s), 5.05 (2H, s), 6.64 (1H, dd, J = 8.0 and 2.3Hz), 6.69-6.78 (2H, m), 6.96 (1H, t, J = 2.3Hz), 7.11 (2H, d, J = 8.3Hz), 7.23 (1H, t, J = 8.3Hz), 7.32 (2H, d, J = 8.3Hz), 7.37 (5H, s), 7.83 (1H, dd, J = 8.9 and 7.7Hz), 8.59 (1H, dd, J = 11.9 and 2.6Hz), 11.01 (1H, brs).	76
10	125	322	2.27 (3H, s), 3.79 (2H, s), 5.05 (2H, s), 6.63 (1H, dd, J = 8.3 and 2.3Hz), 6.74 (1H, dd, J = 8.3 and 2.3Hz), 6.94 (1H, t, J = 2.3Hz), 7.09 (2H, d, J = 8.6Hz), 7.20 (1H, t, J = 8.3Hz), 7.24-7.36 (8H, m) 7.65 (1H, s), 8.65 (1H, d, J = 8.6Hz), 10.78 (1H, brs).	63
15	126	282	1.40-1.60 (8H, m), 1.74-1.92 (6H, m), 2.27 (3H, s), 3.74 (2H, s), 4.29 (1H, m), 6.77 (2H, d, J = 9.2Hz), 6.92 (2H, d, J = 8.9Hz), 6.97 (2H, d, J = 8.6Hz), 7.28 (2H, d, J = 8.9Hz), 7.38 (1H, dd, J = 8.6 and 1.7Hz), 7.92 (1H, d, J = 1.7Hz), 8.63 (1H, d, J = 8.6Hz), 10.67 (1H, brs).	93
20	127	260	(*) 1.01 (6H, d, $J = 6.6Hz$ ), 2.05 (1H, quint, $J = 6.6Hz$ ), 3.64 (2H, d,, $J = 6.6Hz$ ), 3.76 (2H, s), 6.81 (2H, d, $J = 9.2Hz$ ), 6.93 (2H, d, $J = 9.2Hz$ ), 6.98 (2H, d, $J = 8.6Hz$ ), 7.09 (1H, t, $J = 7.5Hz$ ), 7.28 (2H, t, $J = 8.3Hz$ ), 7.59 (1H, td, $J = 7.9Hz$ and 1.6Hz), 8.11 (1H, dd, $J = 8.3$ and 1.6Hz), 8.76 (1H, d, $J = 8.3Hz$ ), 11.74 (1H, brs).	46
25	128	324	3.71 (2H, s), 5.06 (2H, s), 6.84 (2H, d, J = 8.58Hz), 6.91-7.24 (10H, m), 7.31 (2H, d, J = 8.58Hz), 7.56 (1H, dd, J = 7.26, 8.25Hz), 7.94 (1H, dd, J = 1.32, 7.92Hz), 8.50 (1H, d, J = 8.25Hz), 11.21 (1H, s), 13.55 (1H, br).	68
30	129	284	$ 1.37-1.80 \ (14H, m), \ 3.77 \ (2H, s), \ 4.48 \ (1H, tt, J = 3.96, 7.92Hz), \ 6.88 \ (2H, d, J = 8.57Hz), \ 7.02 \ (1H, ddd, J = 1.65, 6.27, 8.57Hz), \ 7.12-7.26 \ (4H, m), \ 7.36 \ (2H, d, J = 8.25Hz), \ 7.64 \ (1H, ddd, J = 1.65, \ 7.26, 8.57Hz), \ 8.03 \ (1H, dd, J = 1.65, \ 7.92Hz), \ 8.57 \ (1H, d, J = 8.57Hz), \ 11.27 \ (1H, s), \ 13.66 \ (1H, br). $	63
35	130	258	3.72 (2H, s), $4.30$ (4H, s), $6.91$ (2H, d, $J=8.6Hz$ ), $6.92-7.01$ (8H, m), $7.12$ (1H, dd, $J=7.9Hz$ and $7.3Hz$ ), $7.30$ (1H, t, $J=7.3Hz$ ), $7.32$ (2H, d, $J=8.6Hz$ ), $7.56$ (1H, ddd, $J=8.6Hz$ , $7.3Hz$ , and $1.7Hz$ ), $7.95$ (1H, dd, $J=7.9Hz$ and $1.7Hz$ ), $8.50$ (1H, d, $J=8.6Hz$ ), $11.24$ (1H, brs), $13.50-13.60$ (1H, br).	37
40	131	262	2.52 (1H, ddd, $J = 6.9$ Hz, $6.6$ Hz, and $1.3$ Hz), $2.55$ (1H, ddd, $J = 6.9$ Hz, $6.9$ Hz, and $1.3$ Hz), $3.72$ (2H,s), $3.87$ (3H, s), $3.99$ (2H, t, $J = 6.9$ Hz), $5.11$ (1H, dd, $J = 10.2$ Hz, and $1.7$ Hz), $5.17$ (1H, dd, $J = 17.1$ Hz, and $1.7$ Hz), $5.86-5.90$ (1H, m), $6.90$ (2H, d, $J = 8.6$ Hz), $6.96$ (4H, s), $7.13$ (1H, dd, $J = 7.6$ Hz, and $7.3$ Hz), $7.32$ (2H, d, $J = 8.6$ Hz), $7.57$ (1H, dd, $J = 8.6$ Hz, and $7.6$ Hz), $7.95$ (1H, d, $J = 7.3$ Hz), $8.50$ (1H, d, $J = 8.6$ Hz), $11.13$ (1H, brs), $13.50-13.60$ (1H, br).	100
45	132	256	$0.90\ (6H,t,J=7.3Hz),1.33\text{-}1.50\ (4H,m),1.57\text{-}1.66\ (1H,m),3.72\ (2H,s),3.83\\ (2H,d,J=5.9Hz),6.90\ (2H,d,J=8.6Hz),6.96\ (4H,s),7.13\ (1H,dd,J=7.6Hz),\\ and7.3Hz),7.32\ (2H,d,J=8.6Hz),7.57\ (1H,dd,J=8.6Hz),and7.6Hz),7.95\\ (1H,d,J=7.3Hz),8.50\ (1H,d,J=8.6Hz),11.13\ (1H,brs),13.50\text{-}13.60\ (1H,br).$	80
50	133	270	0.98 (3H, t, J = $7.3$ Hz), $1.45-1.64$ (2H, m), $1.71-1.82$ (2H, m), $3.72$ (2H, s), $3.93$ (2H, t, J = $6.6$ Hz), $6.90$ (2H, d, J = $8.6$ Hz), $6.96$ (4H, s), $7.13$ (1H, dd, J = $7.6$ Hz, and $7.3$ Hz), $7.32$ (2H, d, J = $8.6$ Hz), $7.57$ (1H, dd, J = $8.6$ Hz, and $7.6$ Hz), $7.95$ (1H, d, J = $7.3$ Hz), $8.50$ (1H, d, J = $8.6$ Hz), $11.13$ (1H, brs), $13.50-13.60$ (1H, br).	100

## [Example 134]

Measurement of cytotoxicity using mouse tumorous cells L929

### 5 [Method]

[0119] The cytotoxic activity against tumorous cells was measured according to a Neutral Red assay method [the method described in Journal of Tissue Culture Methodology, vol. 9, p. 7 (1984), Toxicology Letters, vol. 24, p. 119 (1985)]. Namely, 100  $\mu$ L each of L929 cells (5  $\times$  10<sup>4</sup> cells/mL, 10% FCS/RPM1) was added to a 96-well ELISA plate, and the cells were cultured overnight. The test compound at each measuring concentration was dissolved in a DMSO solution and added, and culturing was further continued for 3 days. To the cultured cells, was then added 2.0  $\mu$ L of Neutral Red so as to provide the final concentration of 0.01%. Incubation was conducted at 37°C for 1 hour, and the cell culture supernatant was removed. The resultant cultured cells were washed with 200  $\mu$ L of PBS twice to remove the excessive Neutral Red. To the washed cells, was then added 100  $\mu$ L of a 50% ethanol-1% aqueous acetic acid. The dye incorporated in the cells was extracted, and the amount of the dye was determined by measuring the absorbance at 490 nm. The case where a drug was not added was taken as 100%, and the cytotoxicity was determined at the concentrations of the respective test compounds. The compound concentration and cytotoxicity at each concentration were plotted for each test compound to determine the concentration (LD<sub>50</sub> value) of the test compound manifesting 50% cytotoxicity. The measurements under the same conditions in two sets each were made, and the data were obtained from the average values. Results are shown in the following tables.

Compound No.	LD <sub>50</sub> (μM)		
4	>5		
22	>5		
29	>5		
37	>5		
59	1.6		
66	>5		
115	>5		
129	>5		
130	0.16		
137	0.29		
145	0.30		
153	0.042		
154	>5		
160	0.31		
168	0.22		
176	0.40		
179	0.170		
185	0.65		
197	3.0		
198	0.052		
200	0.46		
206	0.039		
211	0.060		
212	0.078		

### (continued)

Compound No.	LD <sub>50</sub> (μM)
224	0.34
234	0.29
237	3.0
243	1.2
250	2.5
251	0.340
254	0.44
256	0.048
258	0.75
260	0.210
262	0.64
266	0.25
270	0.30
279	0.080
286	0.15
288	0.20
292	0.17
293	1.6
296	>5
314	0.25
316	0.60
330	7.5
331	1.0
332	0.19

### [Example 135]

Carcinostatic activity against human cultured cancer cells

## [Method]

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[0120] Human cultured cancer cells (39 strains) were seeded in a 96-well plate, and a test substance solution (at concentrations of 5 grades diluted 10-fold ranging from 10<sup>-4</sup> M to 10<sup>-8</sup> M) was added on the next day to carry out culturing for two days. The number of grown cells in each plate was measured by colorimetric determination with Sulforhodamine B. The concentration at which the cell growth was inhibited by 50% (Gl<sub>50</sub> value) as compared with the control (without addition of the test substance) was calculated, and the following values (concentrations) were calculated on the basis of the number of cells just before adding the test substance.

[0121] TGI: the concentration at which the growth is inhibited to the reference number of cells (no apparent increase or decrease in number of cells)

[0122]  $LC_{50}$ : the concentration at which the number of cells is decreased to 50% of the reference number of cells (cellulicidal activity)

The following table collectively shows the results of growth inhibition of the compound 206 of the test substance against representative 9 strains of cancer cells in 39 strains.

Compound No.	Cancer cell strain	Gl <sub>50</sub> (μΜ)	TGI (μM)	LC <sub>50</sub> (μΜ)
206	HBC-4	0.51	27	>100
	SF-539	0.36	20	51
	HCT-15	0.066	17	58
	NCI-H460	0.092	12	53
	LOX-IMVI	0.27	6.3	44
	OVCAR-8	0.92	29	89
	RXF-631L	0.27	16	51
	MKN-74	0.38	22	>100
	PC-3	14	30	62

## **Industrial Applicability**

[0123] The cancer remedy of the present invention has a cytotoxic activity on cell strains having a strong growth property, and further has a strong growth inhibitory activity or cytotoxic activity even against human cancer cells. Therefore, the remedy of the present invention can be used as a remedy for cancer.

# 25 Claims

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1. A cancer remedy comprising an anthranilic acid derivative represented by the following formula (1) or a pharmaceutically acceptable salt thereof as an active ingredient:

$$X \cdot A \longrightarrow H \cdot G$$

$$(1)$$

wherein, X represents a group selected from the following formula (2)-1 and formula (2)-2 in the formula (1):

$$R^1$$
  $(2)$   $-1$  and  $R^2$   $(2)$   $-2$ 

wherein,  $R^1$  and  $R^2$  represent each independently a hydrogen atom, a hydroxy group, a trihalomethyl group, an alkoxy group or an alkylthio group comprising a  $C_1$ - $C_{12}$  chain or cyclic hydrocarbon group and an oxy group or a thio group, a  $C_7$ - $C_{11}$  aralkyloxy group wherein an aryl group moiety may be substituted with one or more halogen atoms, methyl groups or methyloxy groups or a  $C_3$ - $C_{10}$  alkenyloxy group which may be substituted with one or more phenyl groups;  $R^4$  and  $R^5$  represent each independently a hydrogen atom, a halogen atom, a  $C_1$ - $C_4$  alkyl group or a  $C_1$ - $C_4$  alkoxy group, in the formula (2)-1 or the formula (2)-2;

A represents a bond; -O-, -S-, -S(=O)-, -S(=O) $_2$ -, -CH $_2$ -, -OCH $_2$ -, -SCH $_2$ -, -C(=O)- or -CH(OR $^6$ )-, wherein, R $^6$  represents a hydrogen atom or a C $_1$ -C $_4$  alkyl group;

Y represents a hydrogen atom, a halogen atom, a nitro group, a nitrile group, an amino group, -COOR<sup>7</sup>, -NHCOR<sup>8</sup> or -NHSO<sub>2</sub>R<sup>9</sup>,

wherein,  $R^7$  represents a hydrogen atom or a  $C_1$ - $C_4$  alkyl group;  $R^8$  and  $R^9$  represent each independently a  $C_1$ - $C_4$  alkyl group;

E represents a bond; -C(=O)  $-CR^{10}R^{11}C(=O)$ - (wherein,  $R^{10}$  and  $R^{11}$  represent each independently a hydrogen atom or a fluorine atom),  $-CH_2CH_2C(=O)$ - or -CH=CHC(=O)-;

G represents a hydrogen atom, a hydroxy group,  $-SO_2NH_2$ ,  $-COOR^3$ , (wherein,  $R^3$  represents a hydrogen atom or a  $C_1$ - $C_4$  alkyl group), -CN or a tetrazol-5-yl group; and

Z represents a hydrogen atom, a halogen atom, a nitro group or a methyl group.

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- 2. The cancer remedy according to claim 1, wherein G represents -COOR<sup>3</sup> (wherein R<sup>3</sup> represents a hydrogen atom or a C<sub>1</sub>-C<sub>4</sub> alkyl group) or a tetrazol-5-yl group, in the formula (1).
- 3. The cancer remedy according to claim 1 or 2, wherein R<sup>1</sup> is located in the 6-position with respect to the group A (2-position) on the naphthalene ring in the formula (2)-1.
  - 4. The cancer remedy according to claim 1 or 2, wherein, R<sup>2</sup> is located in the 4-position with respect to the group A on the benzene ring in the formula (2)-2.
- 5. The cancer remedy according to any of claims 1 to 4, wherein R<sup>4</sup> and R<sup>5</sup> represent each a hydrogen atom in the formula (2)-2.
  - 6. The cancer remedy according to any of claims 1 to 5, wherein R¹ or R² represents a hydrogen atom; a hydroxy group; an alkoxy group comprising a C₁-C₁₂ chain or cyclic hydrocarbon group and an oxy group; a C₃-C₁₀ alkenyloxy group which may be substituted with one or more phenyl groups; a benzyloxy group, a phenylpropyloxy group or a naphthylmethyloxy group, in the formula (2)-1 or (2)-2.
  - 7. The cancer remedy according to any of claims 1 to 6, wherein A represents -O- or -S- in the formula (1).
- 30 **8.** The cancer remedy according to any of claims 1 to 7, wherein E represents -C(=O)- or -CH<sub>2</sub>C(=O)- in the formula (1).
  - 9. The cancer remedy according to any of claims 1 to 8, wherein the bond A and the bond E are located in the parapositions in the benzene ring substituted with the group Y, in the formula (1).
  - 10. The cancer remedy according to any of claims 1 to 9, wherein Y represents a hydrogen atom, a halogen atom, a nitro group or a nitrile group, in the formula (1).

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## INTERNATIONAL SEARCH REPORT

International application No.
PCT/JP01/01090

A. CLASS Int.	SIFICATION OF SUBJECT MATTER Cl' A61K31/196, 31/215, A61P3	5/00					
According to	According to International Patent Classification (IPC) or to both national classification and IPC						
	SEARCHED						
	ocumentation searched (classification system followed		ools)				
Int.	Cl <sup>7</sup> A61K31/196, 31/215, A61P39	5/00					
				<del> </del>			
	Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched						
	ata base consulted during the international search (nam		iere practicable, sea	rch terms used)			
CAPI	us(stn), medline(stn), embase(stn	")					
C. DOCUI	MENTS CONSIDERED TO BE RELEVANT						
Category®	Citation of document, with indication, where ap	propriate, of the relev	Relevant to claim No.				
Х	WO, 00/05198, A1 (TEIJIN LIMITE 03 February, 2000 (03.02.00), (no family)	ED),		1-10			
	& Database CAPLUS on STN, AME (ACS),(Columbus, OH, USA),DN.13		AL SOCIETY				
A	WO, 95/32943, A1 (TEIJIN LIMITE 07 December, 1995 (07.12.95)	1-10					
i	& EP, 763523, A1 & US, 5945						
	& Database CAPLUS on STN, AME (ACS), (Columbus, OH, USA), DN.12		AL SUCIETY				
A	WO, 97/19910, A1 (TEIJIN LIMITE 05 June, 1997 (05.06.97)	ED),		1-10			
	& EP, 806412, A1 & US, 5808; & Database CAPLUS on STN, AME		AL SOCIETY				
	(ACS), (Columbus, OH, USA), DN.12						
	1 of Day C						
	documents are listed in the continuation of Box C.	See patent fami					
"A" docume	categories of cited documents: nt defining the general state of the art which is not	priority date and	not in conflict with th	mational filing date or e application but cited to			
	ed to be of particular relevance ocument but published on or after the international filing	understand the principle or theory underlying the invention  "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive					
"L" docume cited to	nt which may throw doubts on priority claim(s) or which is establish the publication date of another citation or other	step when the do	cument is taken alone icular relevance; the c	laimed invention cannot be			
"O" docume	eason (as specified) nt referring to an oral disclosure, use, exhibition or other	combined with or	ne or more other such				
	nt published prior to the international filing date but later priority date claimed		g obvious to a person er of the same patent f				
	ctual completion of the international search pril, 2001 (03.04.01)	Date of mailing of the international search report 17 April, 2001 (17.04.01)					
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